





# **Crystallization process development of Febuxostat most stable** polymorph and of a soluble salt thereof

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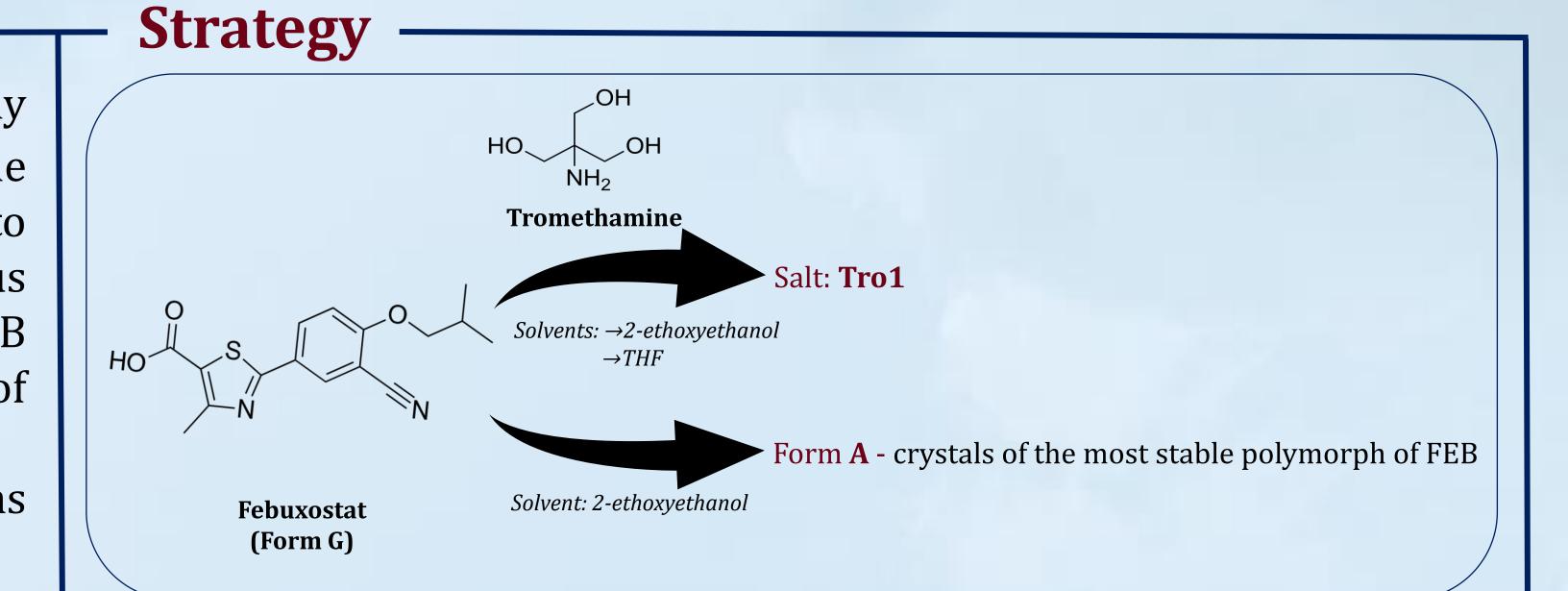
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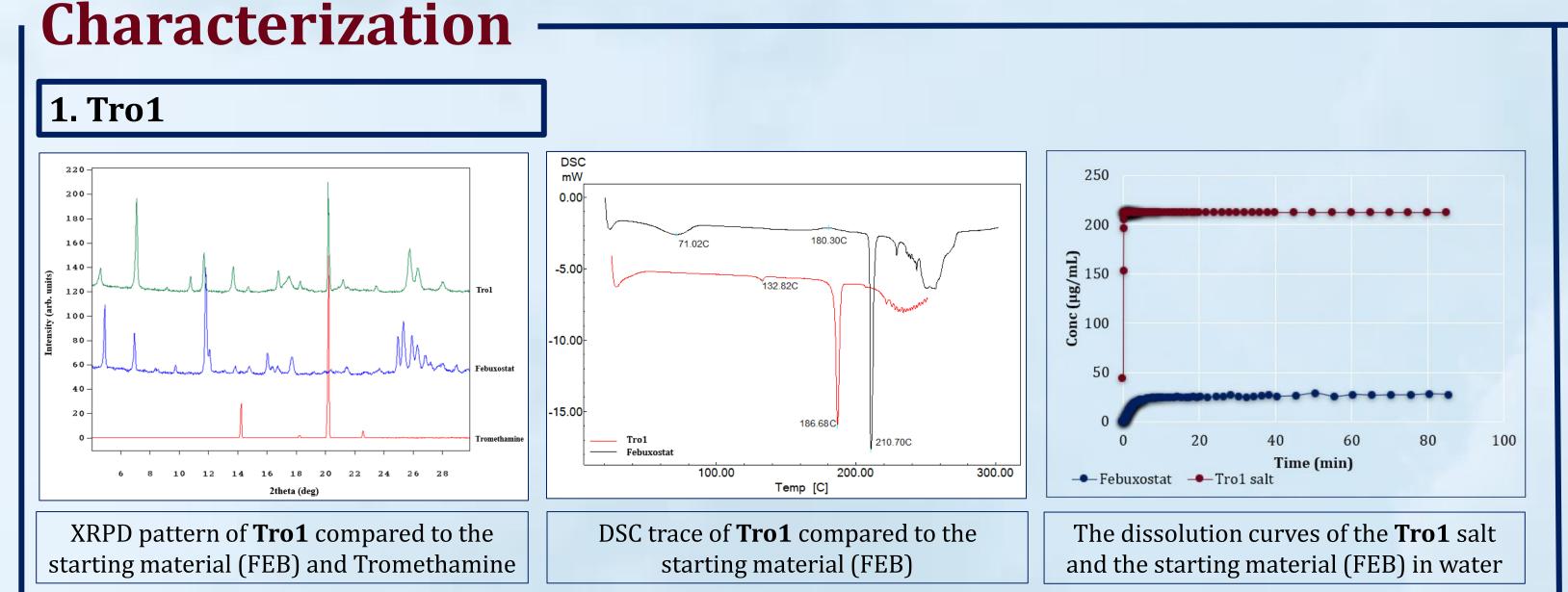
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## Introduction

**H**ebuxostat (FEB) is an active pharmaceutical ingredient (API), poorly water-soluble and therefore poorly bioavailable. FEB is used for the treatment of hyperuricemia in gout, as inhibitor of xanthine oxidase, to reduce uric acid production. With the aim of improving the aqueous solubility, we investigated the crystallization process of the FEB tromethamine salt (Tro1). In addition, we were able to grow crystals of the most stable polymorph of FEB, designated as form A in the literature. The crystallization process development of the Tro1 and form A was performed in a controlled manner using the Crystal16<sup>m</sup>. The crystallization process parameters were established by determining the Meta-stable Zone Width (MSZW) while performing two thermal cycles in a suitable solvent system (2-ethoxyethanol) using a broad temperature range (5 - 90°C) and different concentrations. The formation of Tro1 and form A was evidenced by X-ray powder diffraction. We can conclude that Tro1 can be reliably crystallized in 2-ethoxyethanol by using high starting concentrations (> 200 mg/mL). Furthermore, crystals of form A were successfully grown in the same solvent at concentrations above 100 mg/mL. Crystal structure determination of form A will be attempted in a future work.

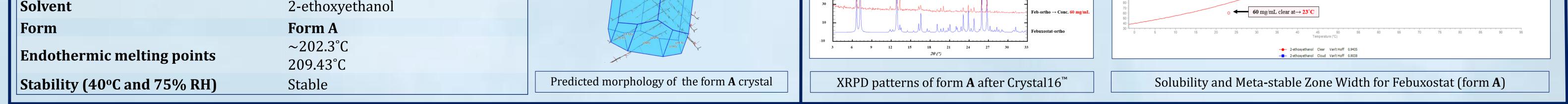






#### **Crystallization using Crystal16**<sup>™</sup>

			1. Tro1			
<b>1. Tro1</b>	DSC	250	Concentration of FEB (mg/mL)	Ratio Feb:Tro	Solvent	<b>Observations after analysis at Crystal16<sup>TM</sup></b> Form by XRPD
200- 180- 160- 140- (Sjim 120- ti to - 100- Trol	0.00 71.02C 180.30C -5.00 132.82C	200 150 100	90 120 240 360	1:1	2-ethoxyethanol	Clear solutionTro1Clear solutionTro1SlurryTro1SlurryTro1SlurryTro1
y       80 -       60 -       60 -       60 -       60 -       Febuxostat         40 -       20 -       0       7       Febuxostat       Febuxostat         0       0       0       7       Febuxostat       Febuxostat         6       8       10       12       14       16       18       20       22       24       26       28         2       2       14       16       18       20       22       24       26       28         2       2       12       14       16       18       20       22       24       26       28	-10.00 -15.00 -15.00 -15.00 -10.00 -186.68C 210.70C Febuxostat 100.00 Temp [C]	<b>5</b> <b>5</b> <b>6</b> <b>7</b> <b>7</b> <b>7</b> <b>7</b> <b>7</b> <b>7</b> <b>7</b> <b>7</b>	200 180 160 140 140 100 100 60 100 100 100 100 100	o1_240.xy	240 mg/mL	
XRPD pattern of <b>Tro1</b> compared to the starting material (FEB) and Tromethamine	DSC trace of <b>Tro1</b> compared to the starting material (FEB)	The dissolution curves of the <b>Tro1</b> salt and the starting material (FEB) in water	$\begin{array}{c} 40 \\ 20 \\ 0 \\ 3 \\ 6 \\ 9 \\ 12 \\ 15 \\ 18 \\ 21 \\ 24 \\ 2\theta \left( {\rm e} \right) \end{array}$	$\mathbf{Tro1} \rightarrow \mathbf{Conc.}$ $\mathbf{Tro1}$ $\mathbf{Tro1}$ $\mathbf{Tro1}$ $\mathbf{Tro1}$	90 mg/mL	120 mg/mL 240 mg/mL 360 mg/mL
Microscope images of Tro1	Starting material       Febuxostat (Form G)         Counter-ion       Tromethamine         Ration (FEB:Tro)       1:1 $\checkmark$ grinding $\checkmark$ grinding         experiment $\checkmark$ slow cooling- evaporative         2-ethoxyethanol       THF         New salt       Tro1         Endothermic melting points       132.82°C 186.68°C       Crystal/5         Solubility in Water       212.11 µg/mL         Stability (40°C and 75% RH)       Stable					
2. Febuxostat – Form A			2. Febuxostat – Form	A		
250 - 200 - (150 - (150 - 100 - 50 -	DSC mW 0.00 -5.00 10.00 -5.00		Concentration of FEB (mg/mL)           60           120           240           360	Solvent 2-ethoxyetha	Clears	after analysis at Crystal16 <sup>™</sup> Form by XRPDClear solutionForm Asolution + precipitate↓Form ASlurryForm ASlurryForm A
$\begin{array}{c} & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ \end{array}$	Heat -124.00 J/g 100.00 200.00 300.00 Temp [C]				390 300 370 360 340 330 310	$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$
XRPD pattern of form A compared to form G       DSC trace of form A       Microscope images of form A			······································	290 280 270 260 250	$\begin{array}{c} 360 \text{ mg/mL clear at} \rightarrow 87^{\circ}\text{C} \\ \hline \\ \hline \\ \hline \\ \end{array}$	
Starting material	Febuxostat (Form G)		10	optho.xy -	220 10 10 10 10 10 10 10 10 10 10 10 10 10	$\circ \circ \leftarrow 240 \text{ mg/mL clear at} \rightarrow 70^{\circ}\text{C}$
Experiment	✓ slow cooling-evaporative		90 Marine Marin	$\mathbf{Feb-ortho} \rightarrow \mathbf{Conc.}$	160	ted at→ 6.5°C
Temperatures	5°C, 25°C, 60°C		June 100 000 000 000 000 000 000 000 000 00	Feb-ortho $\rightarrow$ Conc. 1 Feb-ortho $\rightarrow$ Conc. 1	130 120	$\circ  \bullet  120 \text{ mg/mL clear at} \rightarrow 47.6^{\circ}\text{C}$
Solvent	2-ethoxvethanol		30 The manual have the set		90 TTTT 80 TTTT 70 TT	



### **Conclusions**

- \* A novel salt and crystal form of FEB were obtained in a controlled manner using the Crystal16<sup>™</sup> platform.
- \* Making use of the integrated transmission technology together with 16 parallel reactors at a volume of 1 mL, the Crystal16<sup>™</sup> easily allowed to assess salt and crystal formation.
- formation of **Tro1** and form **A** can be reliably crystallized in Search The 2-ethoxyethanol by using high starting concentrations: > 200 mg/mL for **Tro1**, respectively > **100 mg/mL** for form **A**.

#### References

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[2] Karimi-Jafari, M., Padrela, L., Walker, G. M., & Croker, D. M., Cryst. Growth Des., 2018, 18(10), 6370-6387. [3] Khalaji, M., Potrzebowski, M. J., & Dudek, M. K. Cryst. Growth Des., 2021, , 21(4), 2301-2314. [4] Li L. Y., Du R. K., Du Y. L., Zhang C. J., Guan S., Dong C. Z., Zhang L., *Crystals*, **2018**, *8* (2), 85. [5] Maddileti D., Jayabun S. K., Nangia A., Cryst. Growth Des., 2013, 13 (7), 3188-3196.

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