

# NEW MAGNETIC POLYMERIC COMPOSITE ELECTRODE MATERIAL FOR AMPEROMETRIC NITRITE SENSOR

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## Abstract

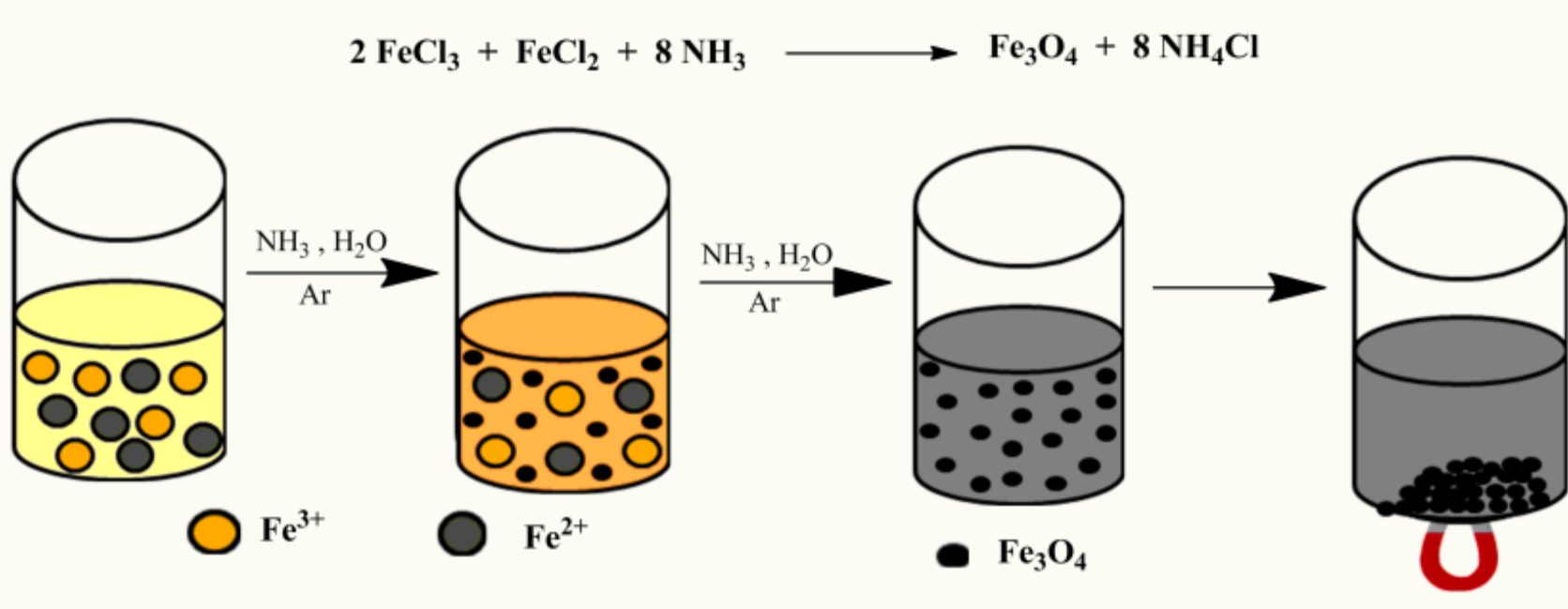
New modified electrode materials, based on a combination of hybrid nanocomposites such as magnetic nanoparticles, inorganic polymer, like silica and organic polymer like poly acrylic acid (pAac) ((MNP-SiO<sub>2</sub> and MNP-SiO<sub>2</sub>-pAac) were studied. The as prepared magnetic polymeric composite materials present high potential applicability as electrochemical sensors for quantitative measurements of nitrite from different kind of real samples.

The electrochemical properties of the synthesized materials were investigated. By electrochemical measurements was proved that this electrode presents electro catalytic effect towards nitrite oxidation and can be used for nitrite detection. By comparing with other electrodes, the electro catalytic effect of MNP-SiO<sub>2</sub>-pAac is better, because the nitrite oxidation peak appears in this case at a potential closer to 0 mV vs. Ag/AgCl/KCl<sub>sat</sub>, which is favourable to avoid interferences in amperometric nitrite detection in real samples

## Synthesis

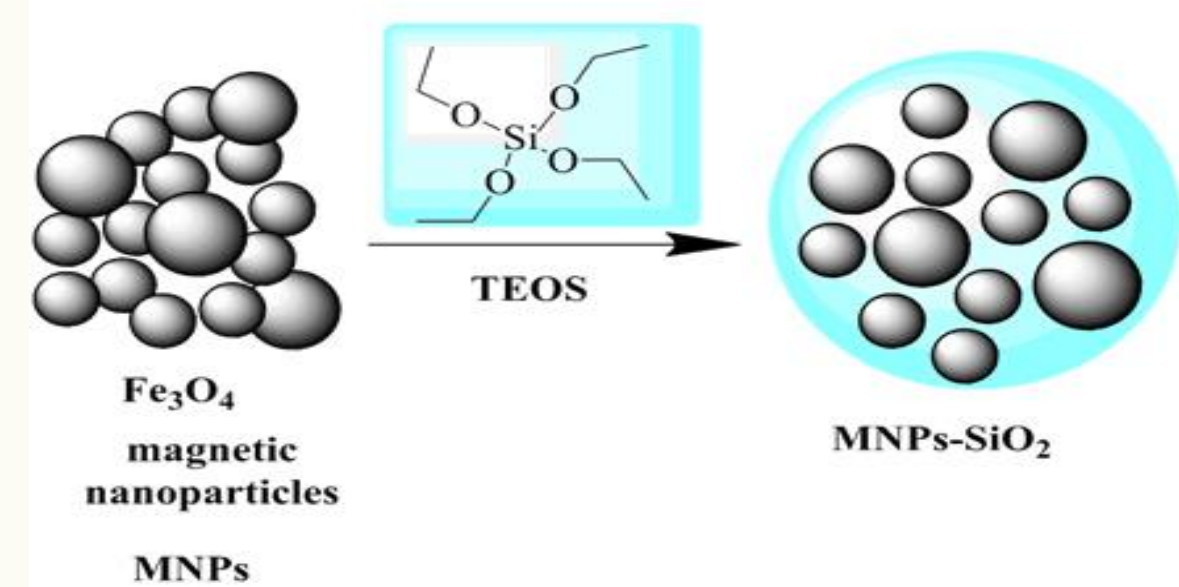
### 1<sup>st</sup> step

Synthesis of Fe<sub>3</sub>O<sub>4</sub> by co-precipitation method (MNPs)



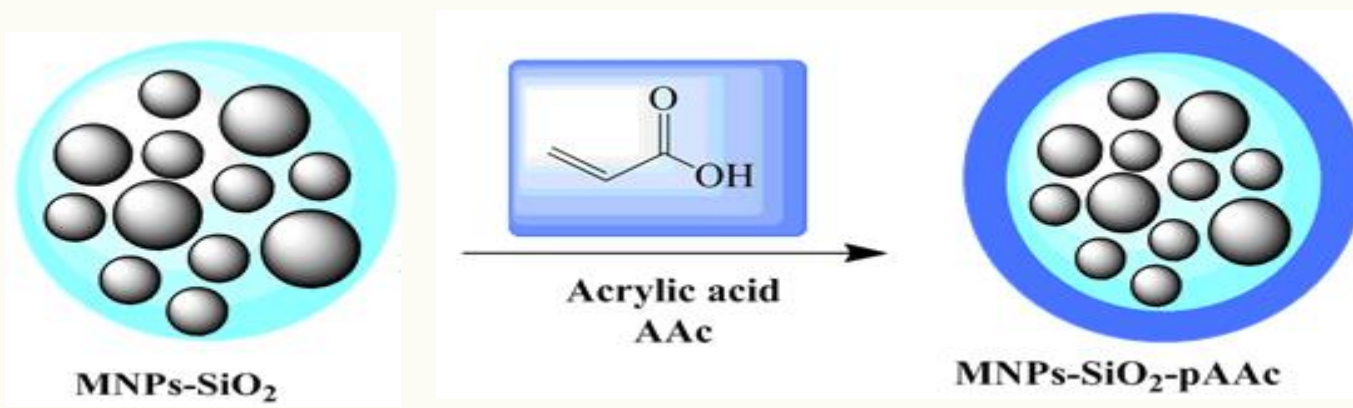
### 2<sup>nd</sup> step

Coating of MNPs with SiO<sub>2</sub> shell (MNP-SiO<sub>2</sub>)



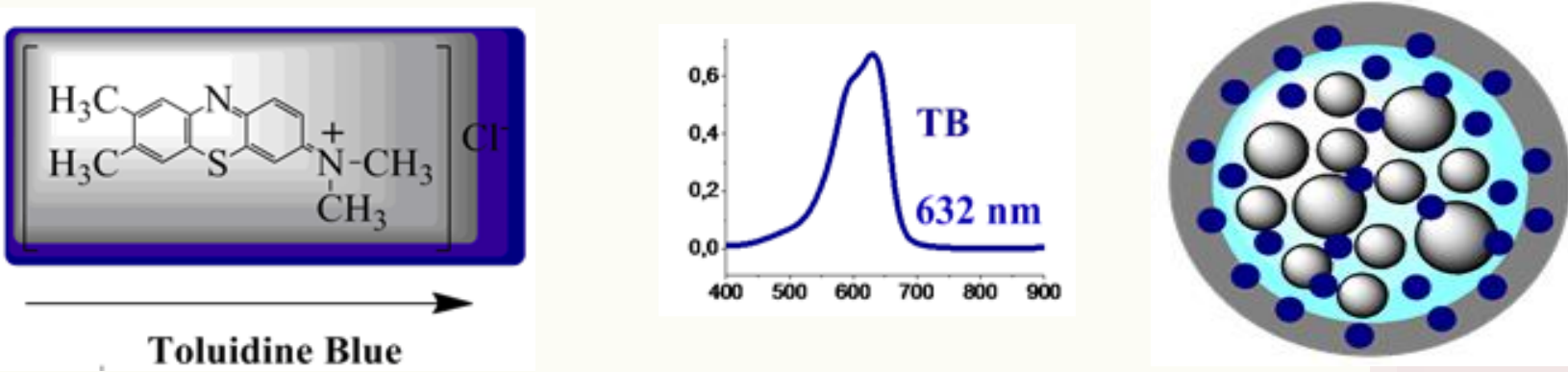
### 3<sup>rd</sup> step

Coating of MNP-SiO<sub>2</sub> with poly acrylic acid (pAac)



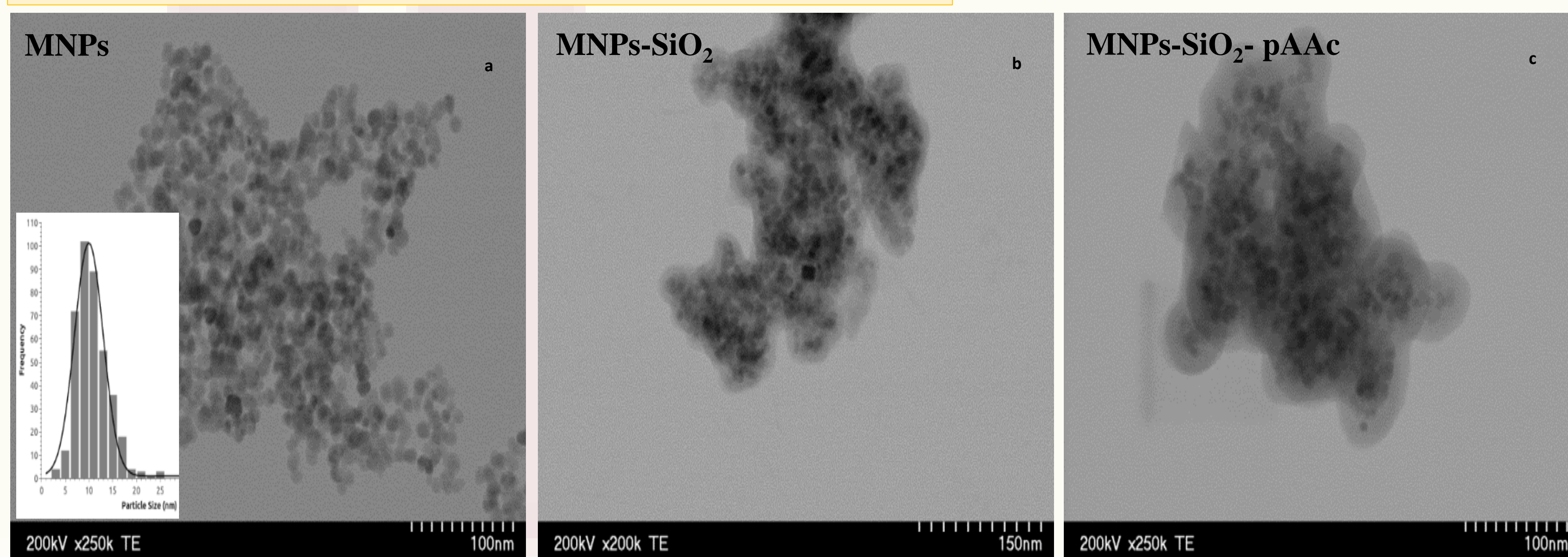
### 4<sup>th</sup> step

TB adsorption on MNP-SiO<sub>2</sub>-pAac

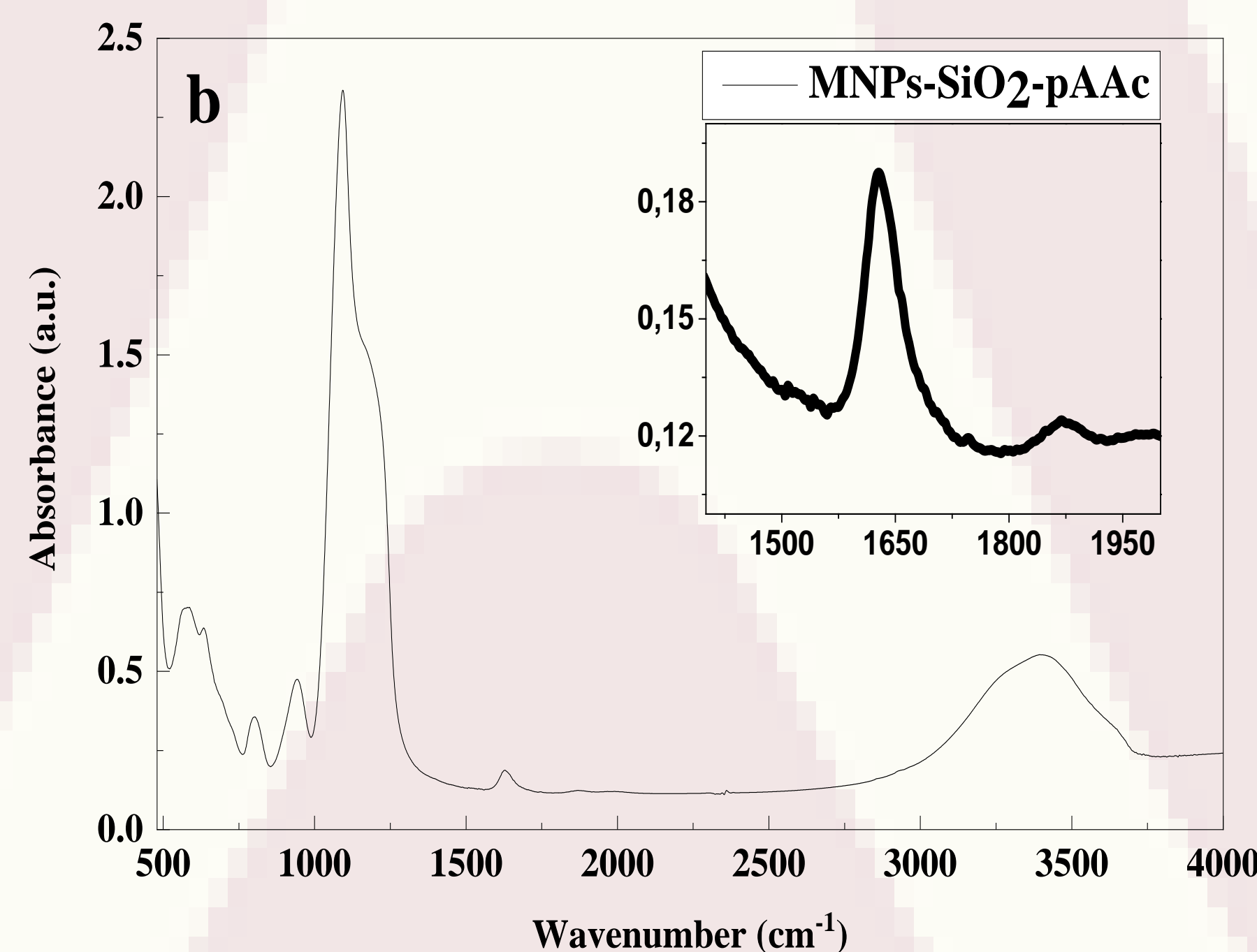


## Characterization

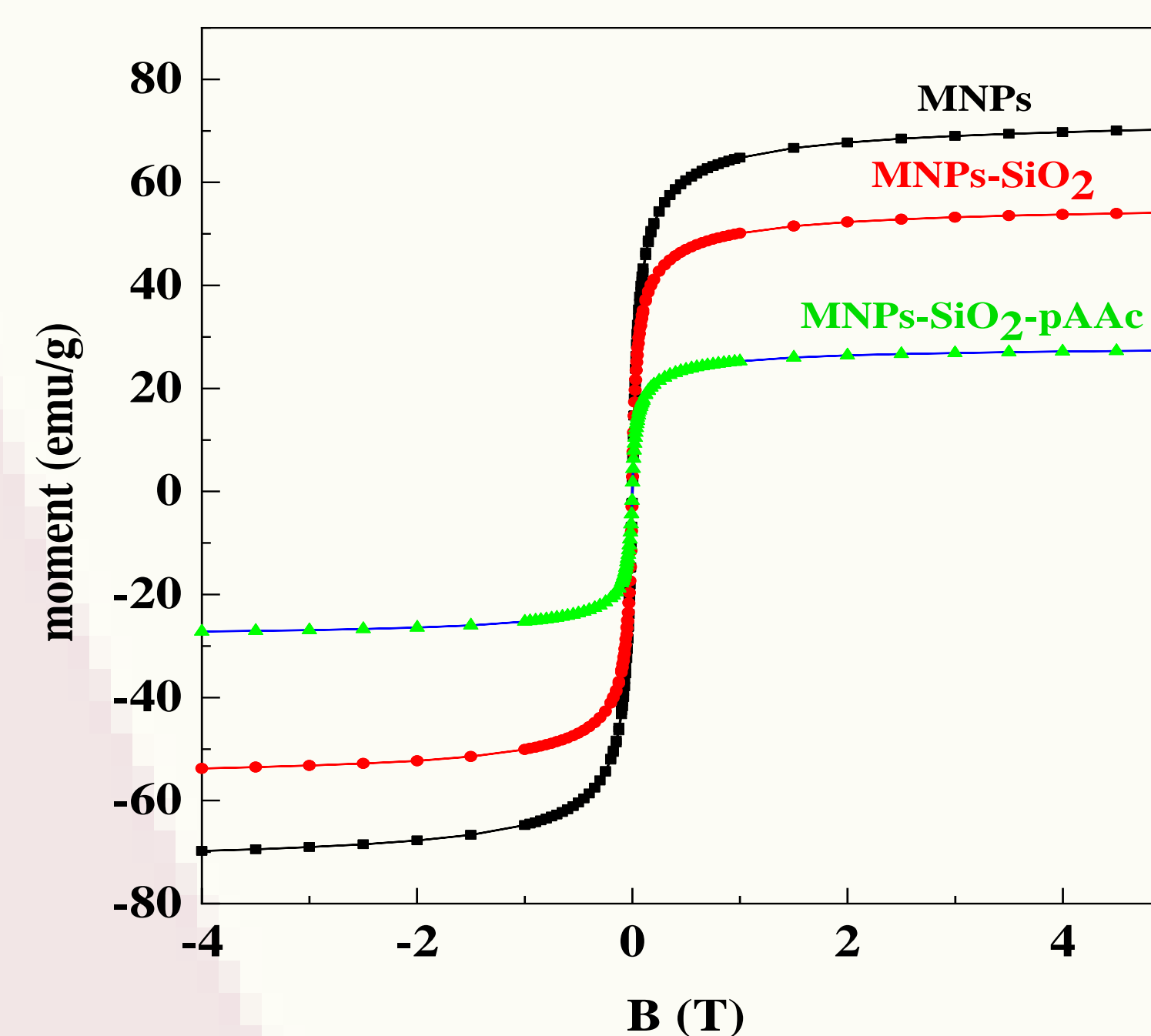
### Morphological characterization by TEM



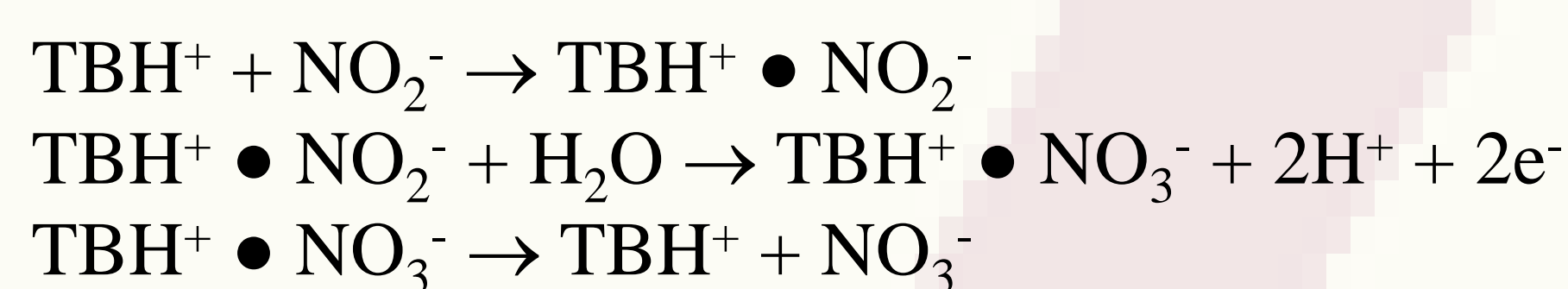
### Structural characterization by FTIR



### Magnetic characterization



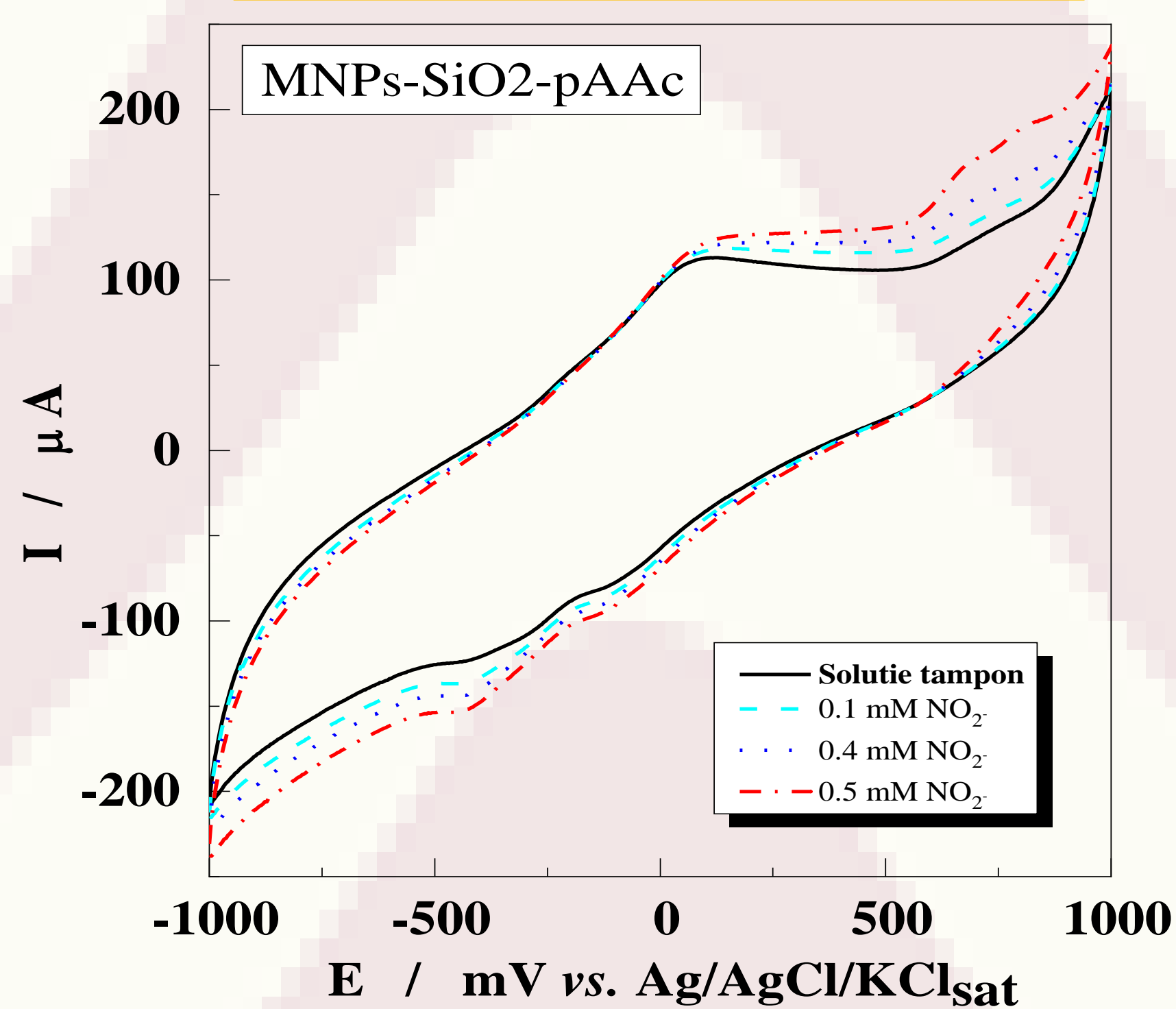
## Electrocatalytic process



## Detection of nitrite in real samples

Water		Soil (solution)			
1	2	1 (0-20 cm/depth)	2 (20-40 cm/depth)	3 (0-20 cm/depth)	4 (20-40 cm/depth)
0.680 (mg/L)	0.704 (mg/L)	1.93 (mg/kg)	1.83 (mg/kg)	2.103 (mg/kg)	1.737 (mg/kg)

## Electrochemical characterization



- ❖ The nitrite oxidation peak appears in this case at a potential closer to +100 mV vs. Ag/AgCl/KCl<sub>sat</sub>
- ❖ The main analytical parameters are: sensitivity: 10.79 mA/M (R/N = 0.999/5); linear domain: 0.08–1 mM; detection limit 20 μM.
- ❖ The possible interference were studied in determination of nitrite. The results showed that 100-fold SO<sub>4</sub><sup>2-</sup>, CO<sub>3</sub><sup>2-</sup>, Cl<sup>-</sup>, NO<sub>3</sub><sup>-</sup>, Na<sup>+</sup>, K<sup>+</sup>, Ca<sup>2+</sup>, Mg<sup>2+</sup> had no effect on the current responses of 0.1 mM nitrite

## Conclusions

- ✓ New modified electrode materials, based on a special combination of hybrid nanocomposites such as MNPs, magnetite (Fe<sub>3</sub>O<sub>4</sub>), inorganic polymer (SiO<sub>2</sub>) and an organic polymer (pAac) were studied.
- ✓ The organic polymer like hydrogel-type polymer (pAac) poly acrylic acid was used to obtain a polymeric shell of cross-linked poly acrylic hydrogels with a favourable hydrogel structure and thicknesses which is perfectly suitable for adsorption of a relatively high concentration of Toluidine Blue (TB).
- ✓ As could be seen from the morphological measurements (TEM/SEM), MNP-SiO<sub>2</sub>-pAac have very well-controlled properties related to the thicknesses, the MNPs are almost individually covered with a thin layer of cross-linked pAac shell which offer a relatively high surface area which is perfectly suitable for adsorption of a relatively high concentration of TB.
- ✓ By electrochemical measurements, it was proved that this electrode presents electrocatalytic effect towards nitrite oxidation and can be used for nitrite detection.
- ✓ The electrocatalytic effect of MNP-SiO<sub>2</sub>-pAac is good because the nitrite oxidation peak appears in this case at a potential closer to +100 mV vs. Ag/AgCl/KCl<sub>sat</sub>
- ✓ The main analytical parameters are: sensitivity: 10.79 mA/M (R/N = 0.999/5); linear domain: 0.08–1 mM. The detection limit was calculated as 20 μM.

## References

1. G.M. Ispas, I. Crăciunescu, S. Porav, R. Turcu and D. Gligor, *Sens. Actuators, A* **276**, 43 (2018). doi:10.1016/j.sna.2018.03.032.
2. M.A. Cîmpean, I. Crăciunescu and D. Gligor, *Mater. Chem. Phys.* **200**, 233 (2017). doi:10.1016/j.matchemphys.2017.07.057.