



# The influence of the vacuum plasma treatment on the cotton surface deposited by $TiO_2$ / $TO_2$ +graphene nanopowder water colloidal dispersions

A. O. Mateescu<sup>1</sup>, G. Mateescu<sup>1</sup>, I. Burducea<sup>1</sup>, P. E Mereuță<sup>1</sup>, C. Socaci, M. C. Rosu<sup>2</sup>

<sup>1</sup>"Horia Hulubei" National Institute for Research and Development in Physics and Nuclear Engineering, 30 Reactorului Street, 077125 Măgurele, Ilfov, Romania

<sup>2</sup>National Institute of Materials Physics, 405A Atomîștilor Street, 077125 Măgurele, Ilfov, Romania

<sup>3</sup>National Institute for Research and Development of Isotopic and Molecular Technologies, 67-103 Donat, 400293 Cluj-Napoca, Romania  
amateescu@niham.nipne.ro

## Introduction

The influence of the vacuum plasma treatment on the deposited surfaces of cotton samples by their immersion in  $TiO_2$  and  $TiO_2$ +graphene water colloidal dispersions

Vacuum plasma treatment

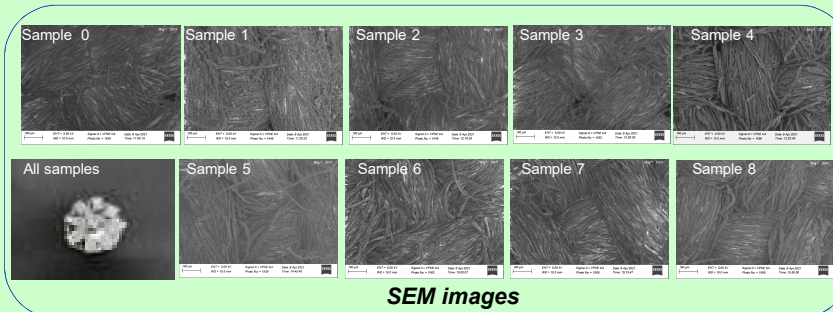
## Experimental

* Sample no.	0	1	2	3	4
Type of Sample	Blank	** Immersed in $TiO_2$ dispersion(1%)	**** Treated in vacuum plasma (air) → Immersed in $TiO_2$ dispersion(1%)	Treated in vacuum plasma (air) → Immersed in $TiO_2$ dispersion(1%) → Treated in vacuum plasma (air)	Immersed in $TiO_2$ dispersion(1%) → Treated in vacuum plasma (air)
Sample no.	5	6	7	8	
Type of Sample	Immersed in $TiO_2$ +Gr (9:1) dispersion(1%)	Treated in vacuum plasma (air) → Immersed in $TiO_2$ +Gr(9:1) dispersion(1%)	Treated in vacuum plasma (air) → Immersed in $TiO_2$ +Gr (9:1) dispersion(1%) → Treated in vacuum plasma (air)	Immersed in $TiO_2$ +Gr (9:1) dispersion(1%) → Treated in vacuum plasma (air)	

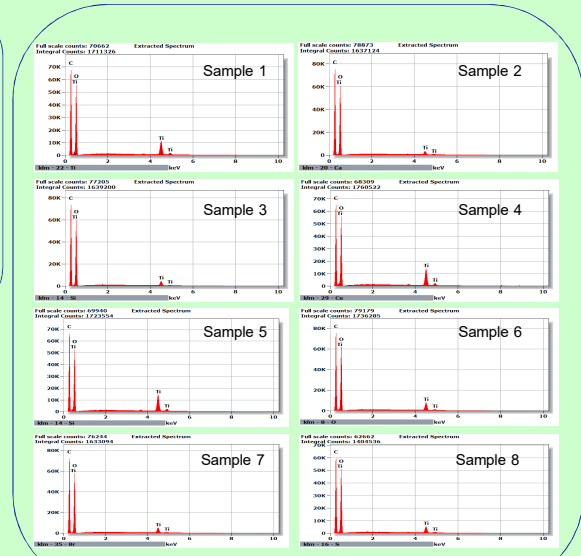
- \* Cotton
- \*\* Distillated water
- \*\*\* In ultrasound bath
- \*\*\*\* 10 minutes - each side



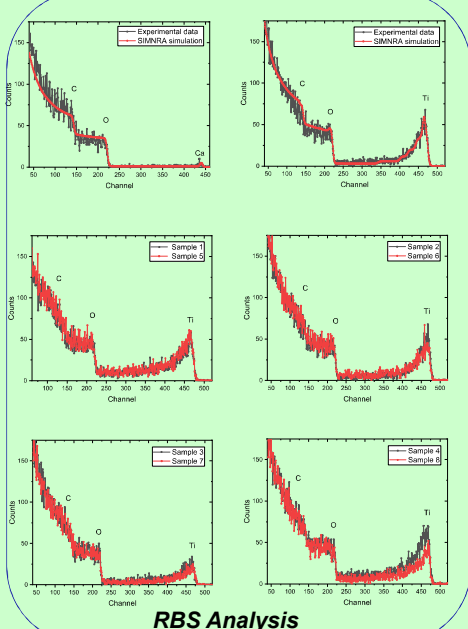
## Results



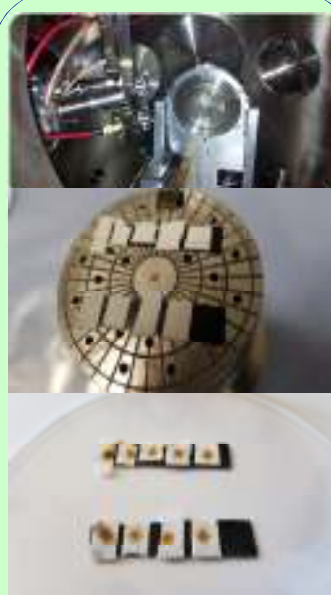
SEM images



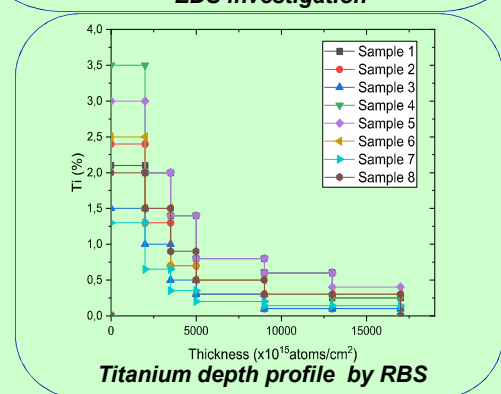
EDS investigation



RBS Analysis



RBS Measurements



Titanium depth profile by RBS

## Conclusions

- By both investigation methods (SEM-EDS and RBS) of the samples that were treated after  $TiO_2$  nanopowder deposition process by vacuum plasma one can observe a low reduction of the  $TiO_2$  nanopowdwer quantity, which means, the low adherent to substrate  $TiO_2$  nanoparticles were removed during this treatment, remaining only high adherent  $TiO_2$  nanopowdwers on the cotton surface.
- Improvement of the  $TiO_2$  nanopowdwers adherence to the substrate is due to the stimulative effect in creating chemical bonds between cotton substrate and  $TiO_2$  nanoparticles as a result of the free radicals from substrate and electrical charged nanoparticles that generate stable chemical bonds by their high reactivity.

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