

Surface interaction studies of novel 2D materials with gram-negative and gram-positive pathogens.

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Abstract

The interaction between 2D-nanoflakes and bacteria in water based physiological liquids is a hot topic in biosciences. In this work we extend the DLVO theory, to the case of 2D-nanoflakes interacting with bacteria cell membranes, both for gram-positive and gram-negative bacteria. We study the role of the bacterial shape, membrane potential and 2D-materials nature. We calculate the interaction distances at equilibrium for different bacterial species and MoS_2 nanomaterials in Water and Cyrene.

Introduction

The interaction of bacteria and nanoparticles¹ and with 2D-nanomaterials² are well studied. But still, there is more investigation needed—for example, studies of the interaction of different bacteria and different solvents. Here, we used DLVO (Derjaguin-Landau-Verwey-Overbeek)³ model to understand how different types of bacteria (Three Gram-positive and Three Gram-negative), interact with 2D-nanomaterials (MoS_2) in two different solvents (Water and Cyrene), by studying surface interaction energy, and, how different characteristics of bacteria (such as ζ -potential and radius) affect their interaction energy.

Methodology

- DLVO model below equations (two main interactions, the attractive van der Waals (V^{VW}) and the repulsive electrostatic (V^{EL})) equation below.
- Six bacteria strains (S.carnosus, S.maltophilia, N.subflava, B.subtilis, S.aureus and E.coli).
- MoS_2 nanoflake in two different green solvents (Cyrene and Water).
- The critical volume r_{crit}^3 (The critical volume around the bacteria where the attraction forces dominate)¹ is calculated by using below equation.

$$V^{tot} = V^{EL} + V^{VW}$$

$$V^{EL} = \frac{\pi\epsilon a_1 a_2 (\zeta_1^2 + \zeta_2^2)}{(a_1 + a_2)} \left\{ \frac{2\zeta_1 \zeta_2}{\zeta_1^2 + \zeta_2^2} \ln \left[\frac{1 + \exp(-kd)}{1 - \exp(-kd)} \right] + \ln[1 - \exp(-2kd)] \right\}$$

$$V^{VW} = -Aa_1 a_2 / 6d(a_1 + a_2)$$

$$r_{crit}^3 = (d_{max} + a_1)^3 - a_1^3$$

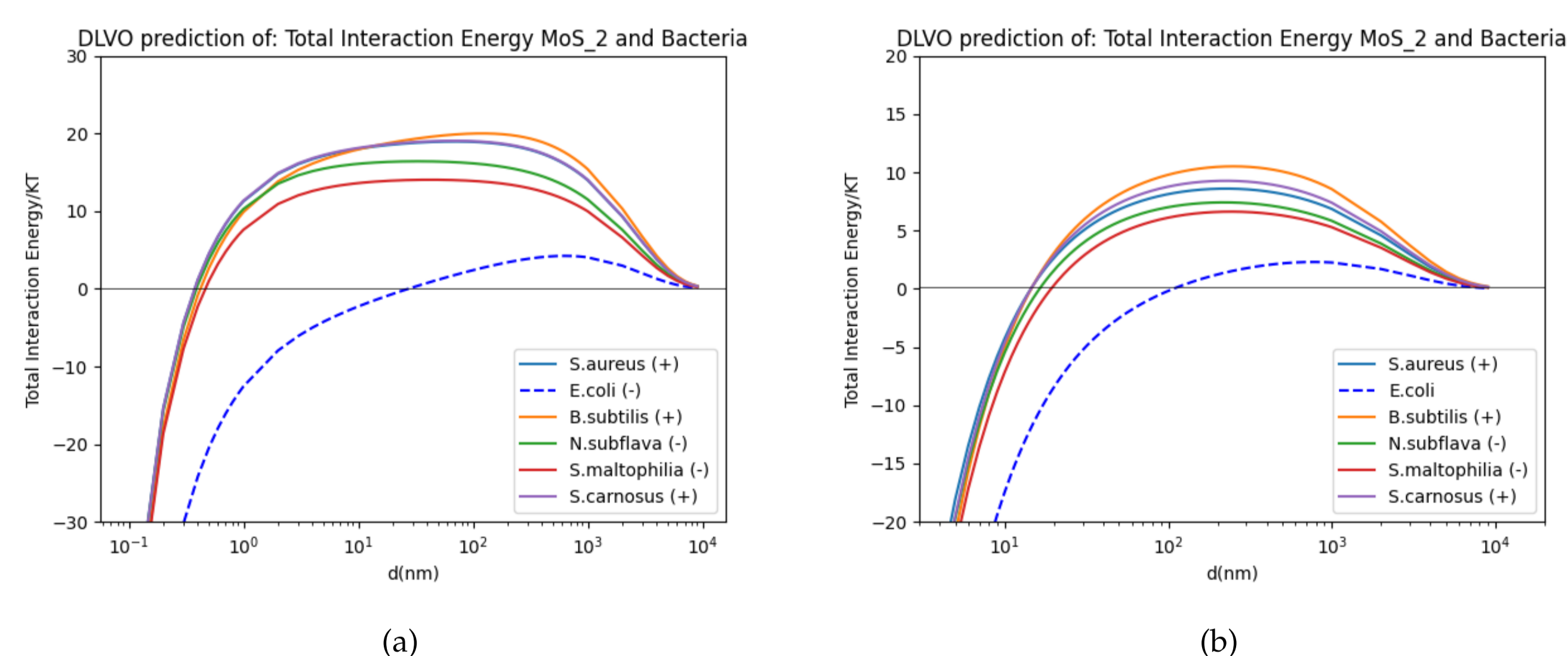


Figure 1: Total Interaction Energy of bacteria- MoS_2 for a) Water b) Cyrene.

	ζ -poten (mV)	a_1 (nm)	d_{water} (nm)	d_{cyrene} nm	V_{water}^{tot}/KT	V_{cyrene}^{tot}/KT	$r_{crit_w}^3$
B.subtilis (G+)	-41.00	520	120.00	249.98	20.00	10.51	12.20
S.carnosus (G+)	-37.00	440	68.00	224.98	19.05	09.27	04.50
S.aureus (G+)	-37.10	360	69.00	224.98	18.96	08.59	03.23
N.subflava (G-)	-30.00	400	33.00	217.98	16.41	07.41	01.72
S.maltophilia (G-)	-26.00	460	43.00	240.98	14.03	06.62	02.99
E.coli (G-)	-12.70	630	629.00	771.98	04.23	02.30	175.00

Table 1: Summary of different bacteria results and characteristic.

Results and Discussion

- The relation between the total interaction energy and spacing (d nm) between the bacteria and MoS_2 nanoflake for both Water and Cyrene Figure 1.
- Higher interaction of bacteria with Water-based solvent than that of Cyrene Figure 2 (a), mainly due to Water higher Hamaker constant.
- The G+ bacteria have higher interaction energy than G- due to the shape of bacteria and different ζ -potential (higher in the case of G+)Figure 1.
- Direct proportionality between the interaction energy and ζ -potential for both bacteria type.
- This pattern of results is consistent with the previous literature¹.
- The radius in the G+ bacteria is directly proportional to energy, While the radius in the G- bacteria is inversely proportional to the energy.
- In Figure 2 (b) we selected a single value for the spacing (d), plotted and compared the V^{vw} and V^{EL} for each solvent.
- The V^{EL} term is dominant for Water and V^{vw} dominant for the case of Cyrene Figure 2(b).

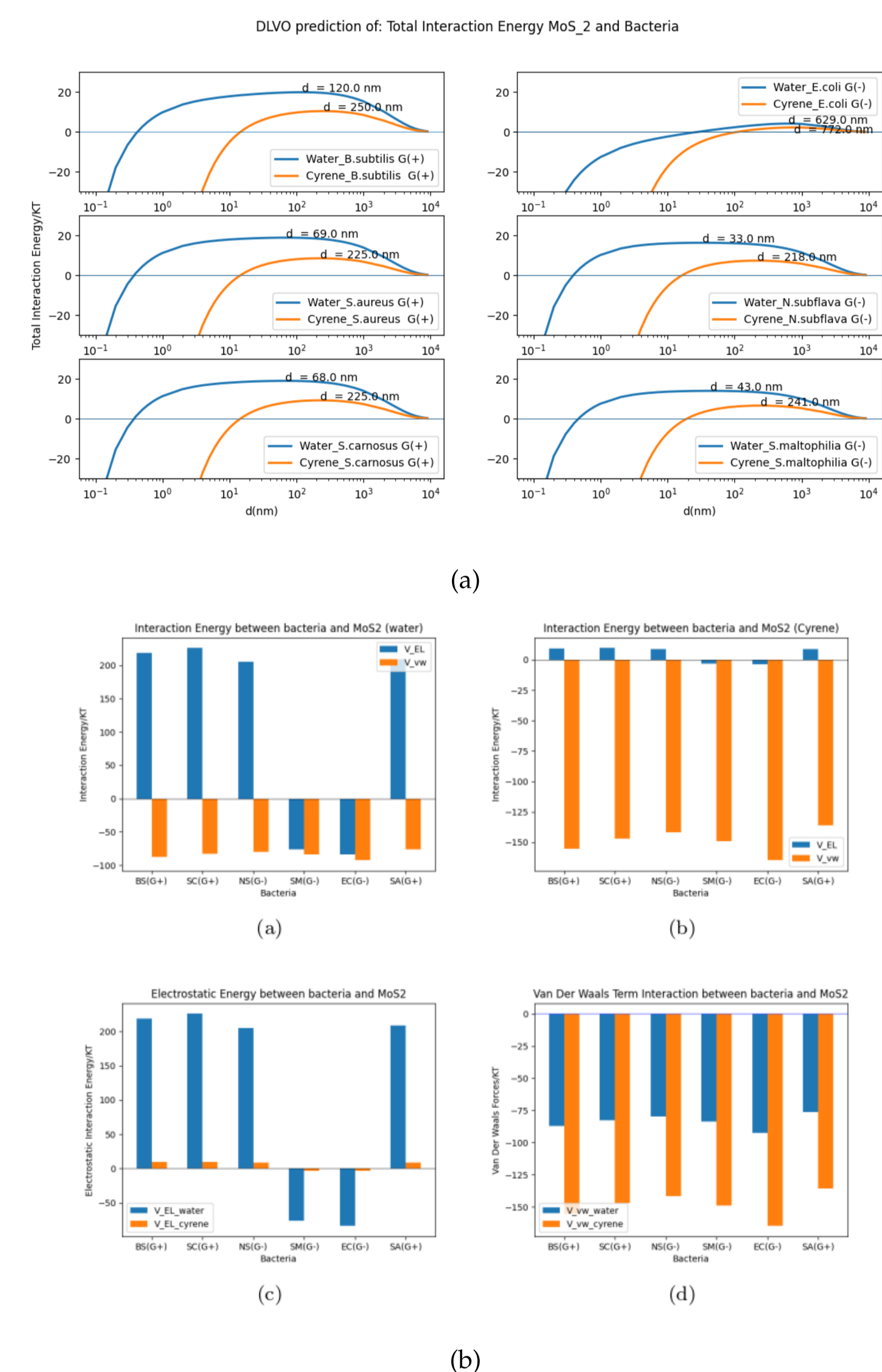


Figure 2: a) Comparison between Water and Cyrene based solvents MoS_2 -bacteria Total Interaction Energy b) V^{EL} and V^{vw} Interaction for Water and Cyrene (Upper panel), V^{EL} and V^{vw} for each solvents. (lower panel)

Conclusion/prospectives

We examined the surface interaction of bacteria and MoS_2 2D-nanoflakes based on two solvents using DLVO model. We found that G+ bacteria have higher interaction energy than G- bacteria. Water-based solvent MoS_2 nanoflake interacts more than Cyrene. We found that as the negative value of ζ -potential increase, interaction energy increase as well. And the critical volume, as well as Electrostatic and Van der Waals, was calculated. Also, in terms of future research, it would be helpful to extend the present findings by examining others solvents and other 2D-nanomaterials.

References

