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1. INTRODUCTION

In modern society, nanomaterials are showing great potential for enabling and improving technologies for water treatment by photocatalysis. In photocatalytic processes reactive oxygen species (ROS) play crucial roles in the degradation of most organic compounds (dyes, antibiotics, etc) in the solution. g-C₃N₄-TiO₂ nanocomposites has attracted much attention due to its functionalities combining the properties of TiO₂ and graphitic carbon nitride (g-C₃N₄) in one single entity. The present work is focused on the photocatalytic properties of g-C₃N₄-TiO₂:Cu nanocomposites obtained by deposition of TiO₂:Cu nanoparticles obtained through a sol-gel process onto the g-C₃N₄ resulted from decomposition of urea. The preparation conditions of these composites are correlated with the need to obtain electron-hole recombination rates as low as possible, thus increasing the efficiency of the photocatalytic process.

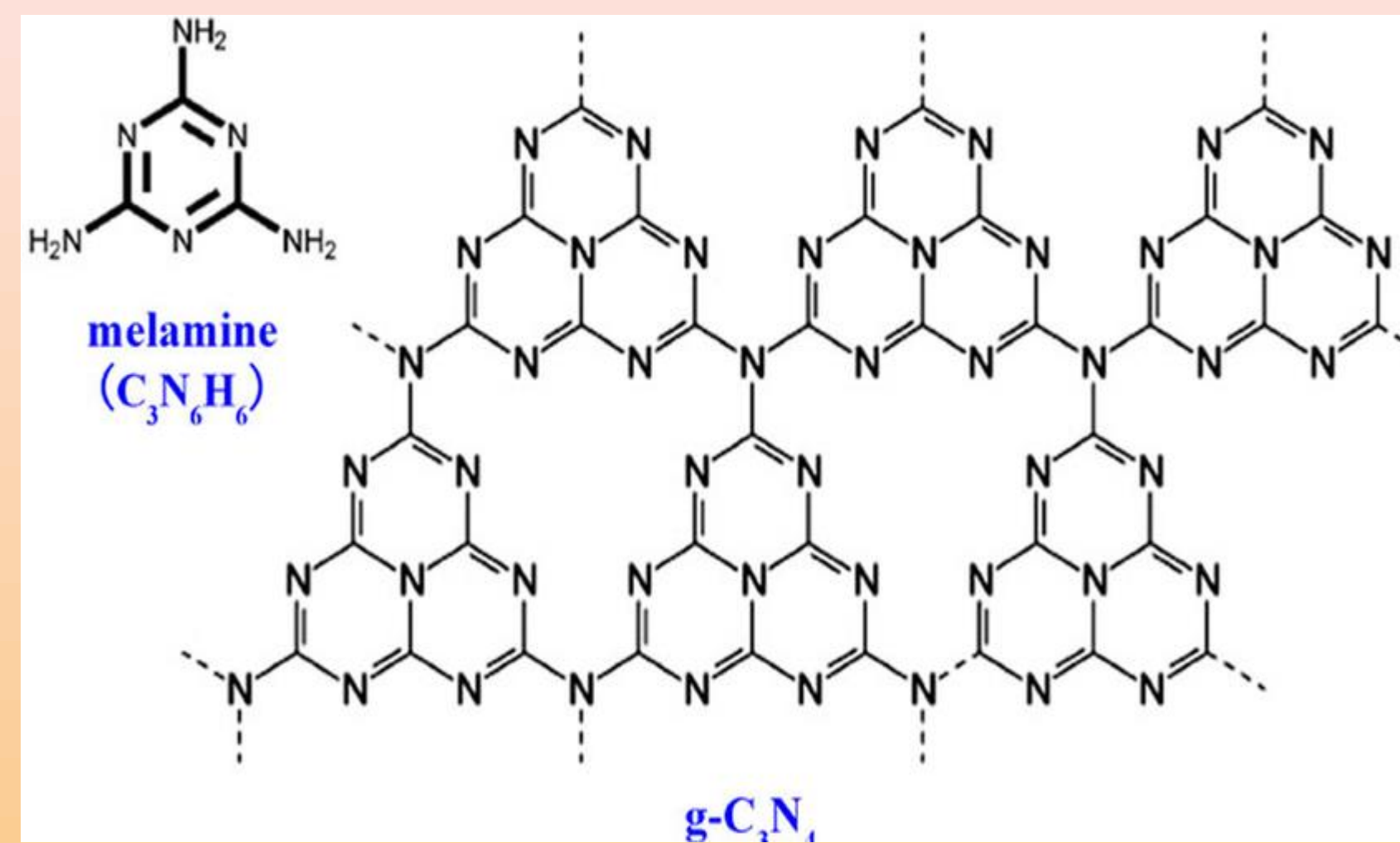
2. EXPERIMENTAL PART

Preparation methods:

g-C₃N₄-TiO₂:CuX% composite nanoparticles were prepared in two stages:

g-C₃N₄ nanosheets –thermal decomposition of urea
TiO₂ nanoparticles-sol-gel, alcoxidic route
Finally, the g-C₃N₄-TiO₂ composite nanoparticles were thermally treated for 2h at 550°C

Different samples with different Cu ions concentrations: g-C₃N₄-TiO₂:CuX% (x= 0- 5mol%)



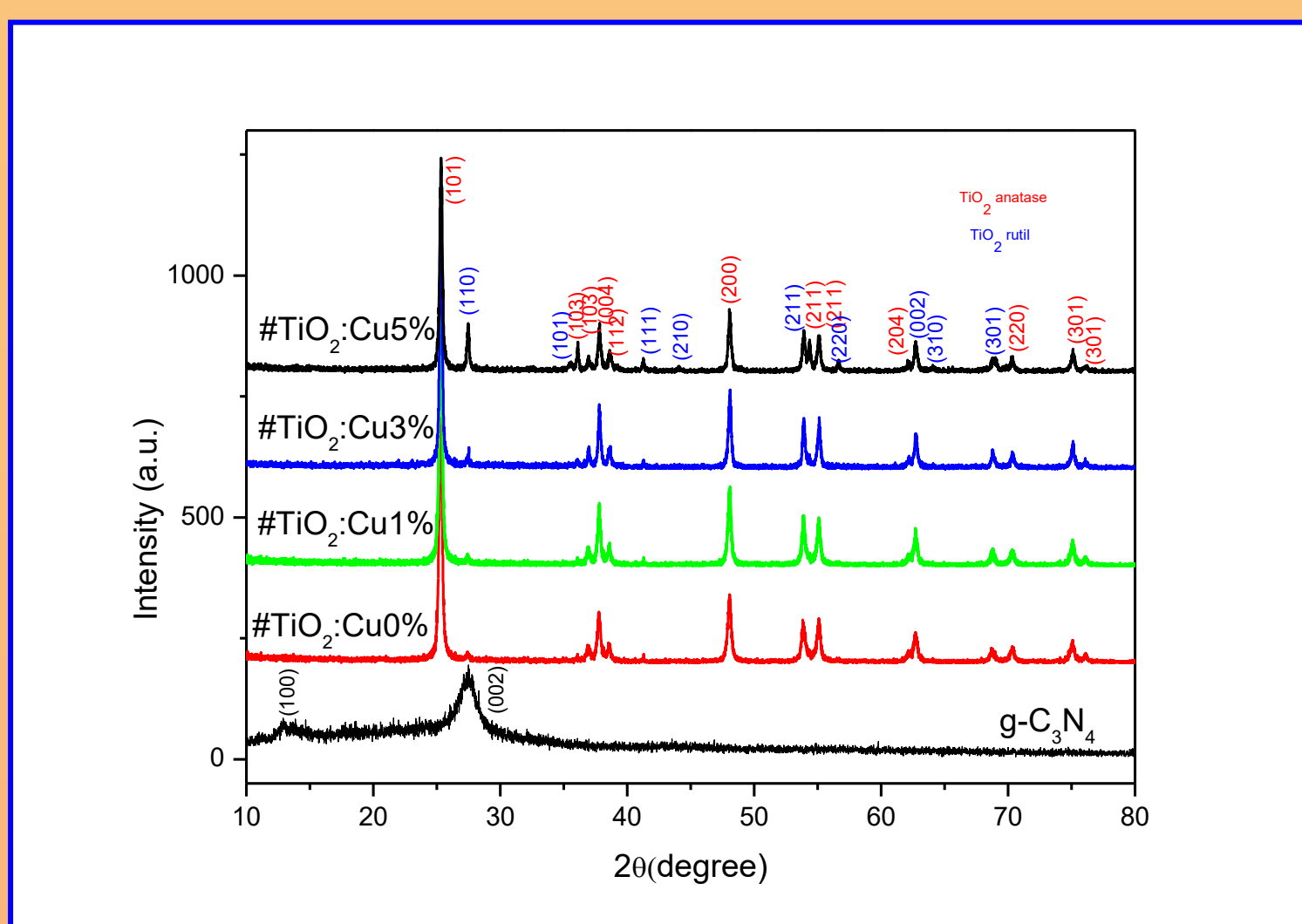
g-C₃N₄ structure

Samples characterization:

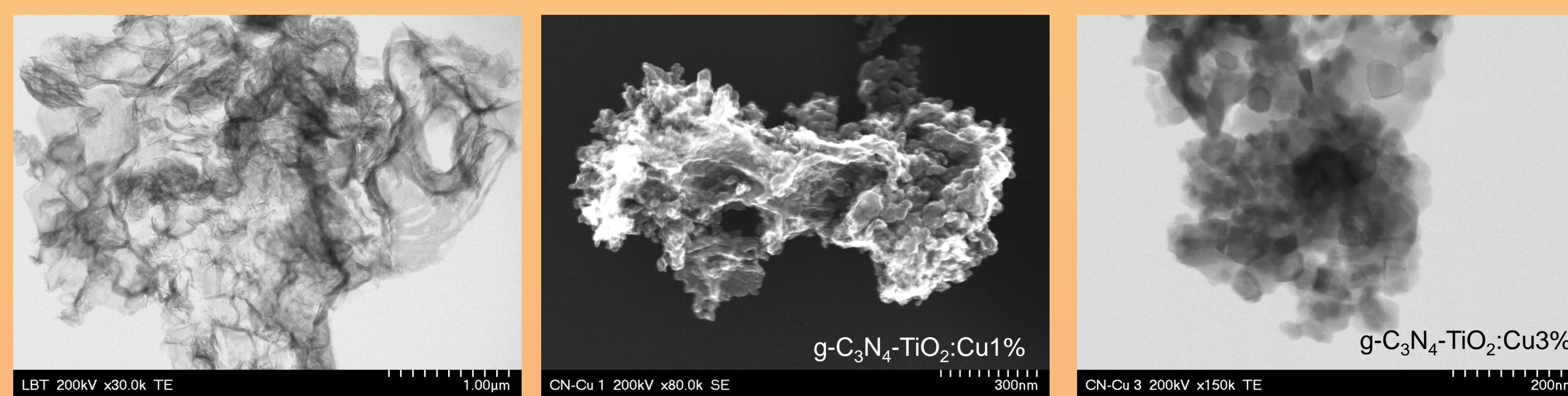
- X-Ray Scattering patterns ⇔ Rigaku - SmartLab automated Multipurpose X-ray Diffractometer
- TEM images ⇔ Hitachi H9000NAR transmission electron microscope
- UV-VIS absorption spectra ⇔ JASCO V570UV- VIS-NIR Spectrophotometeequipped with absolute reflectivity measurement JASCO ARN-475 accessory
- FT-IR spectra ⇔ JASCO 610 Spectrometer (KBr pellets technique).
- ROS generation ⇔ EPR spin trapping technique
- XPS spectra ⇔ SPECS custom built system

3. RESULTS AND DISCUSSION

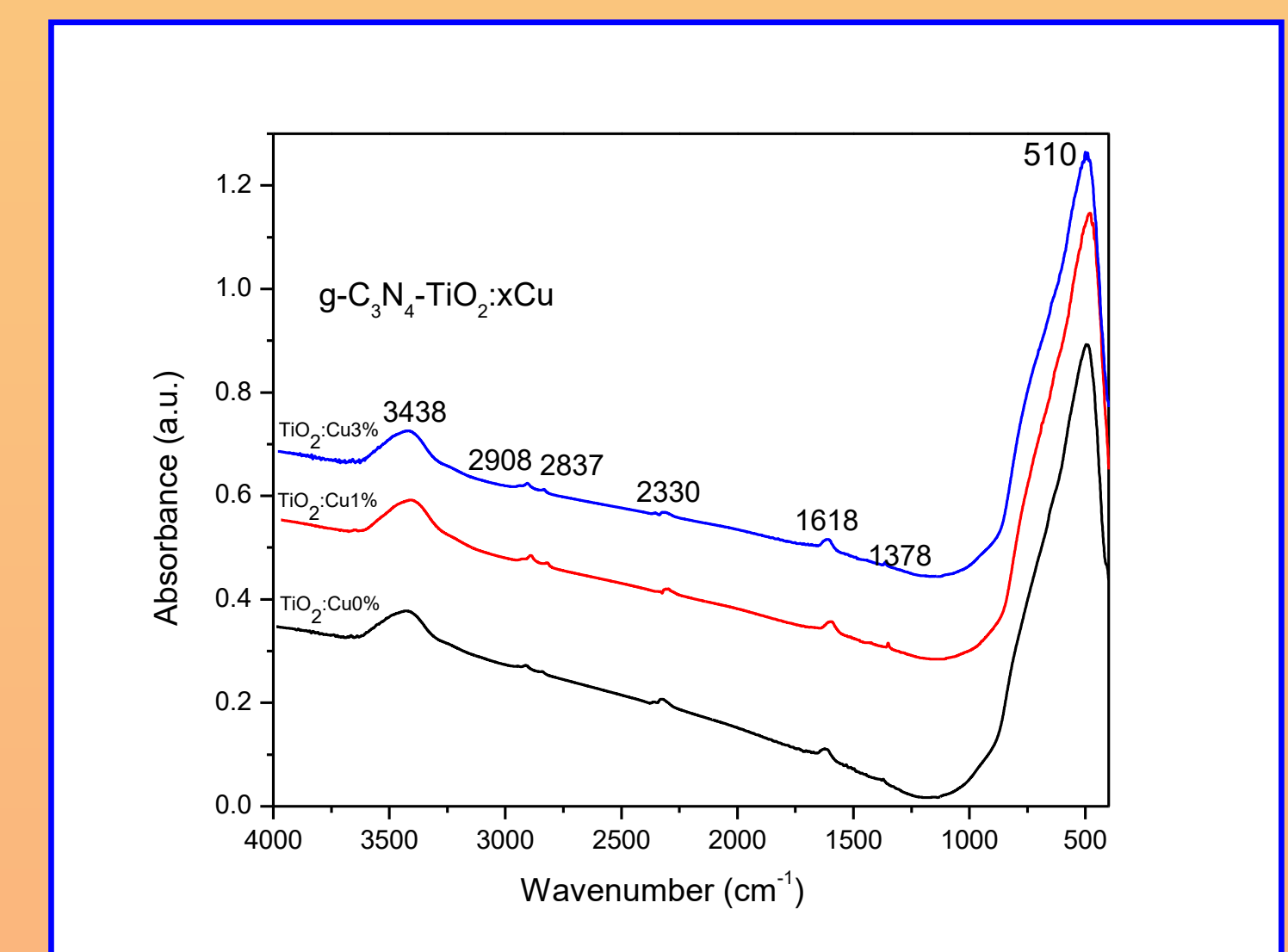
3.1 Structure, morphology and composition



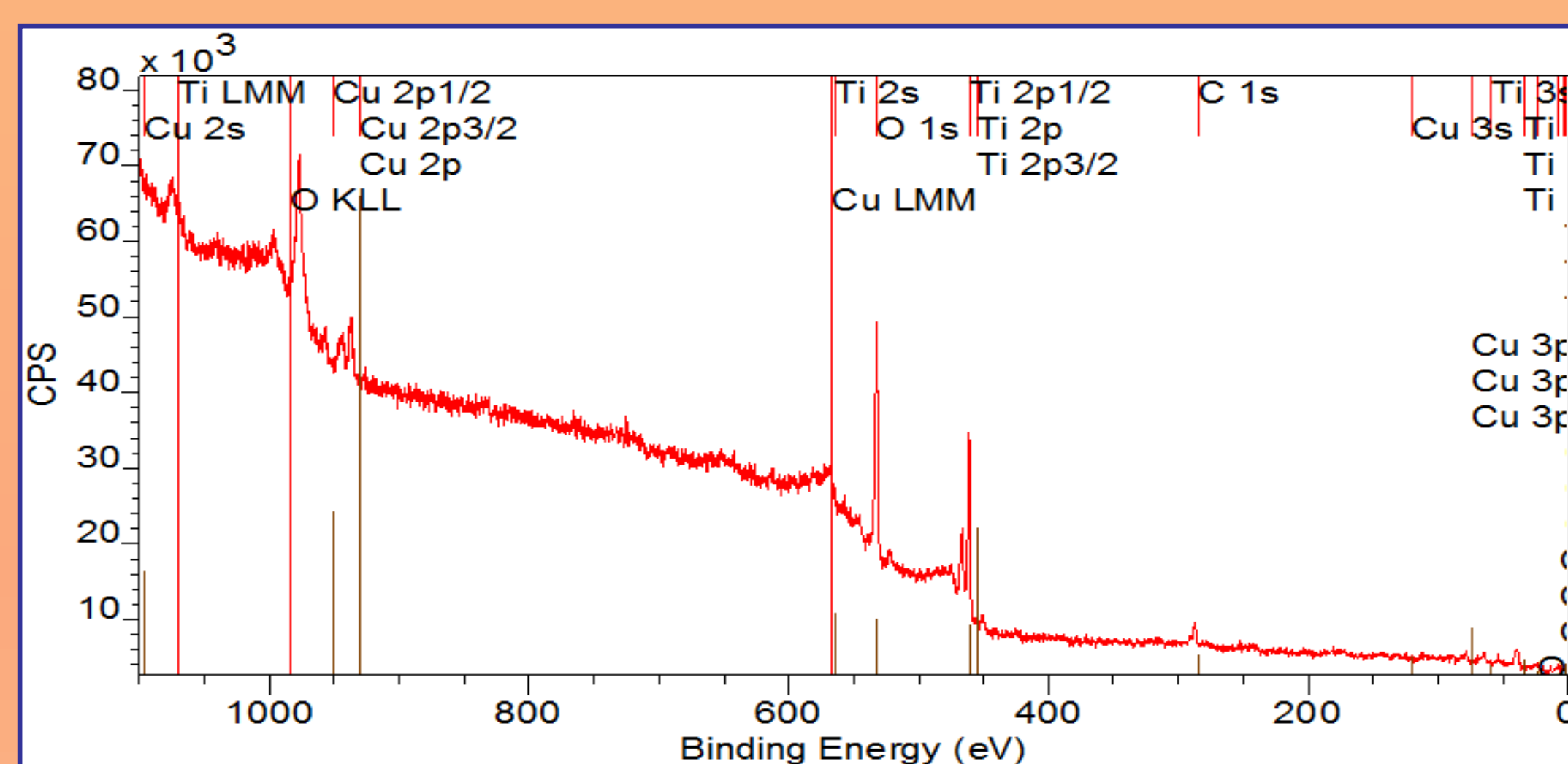
XRD patterns of g-C₃N₄ and g-C₃N₄-TiO₂:CuX% based nanocomposites



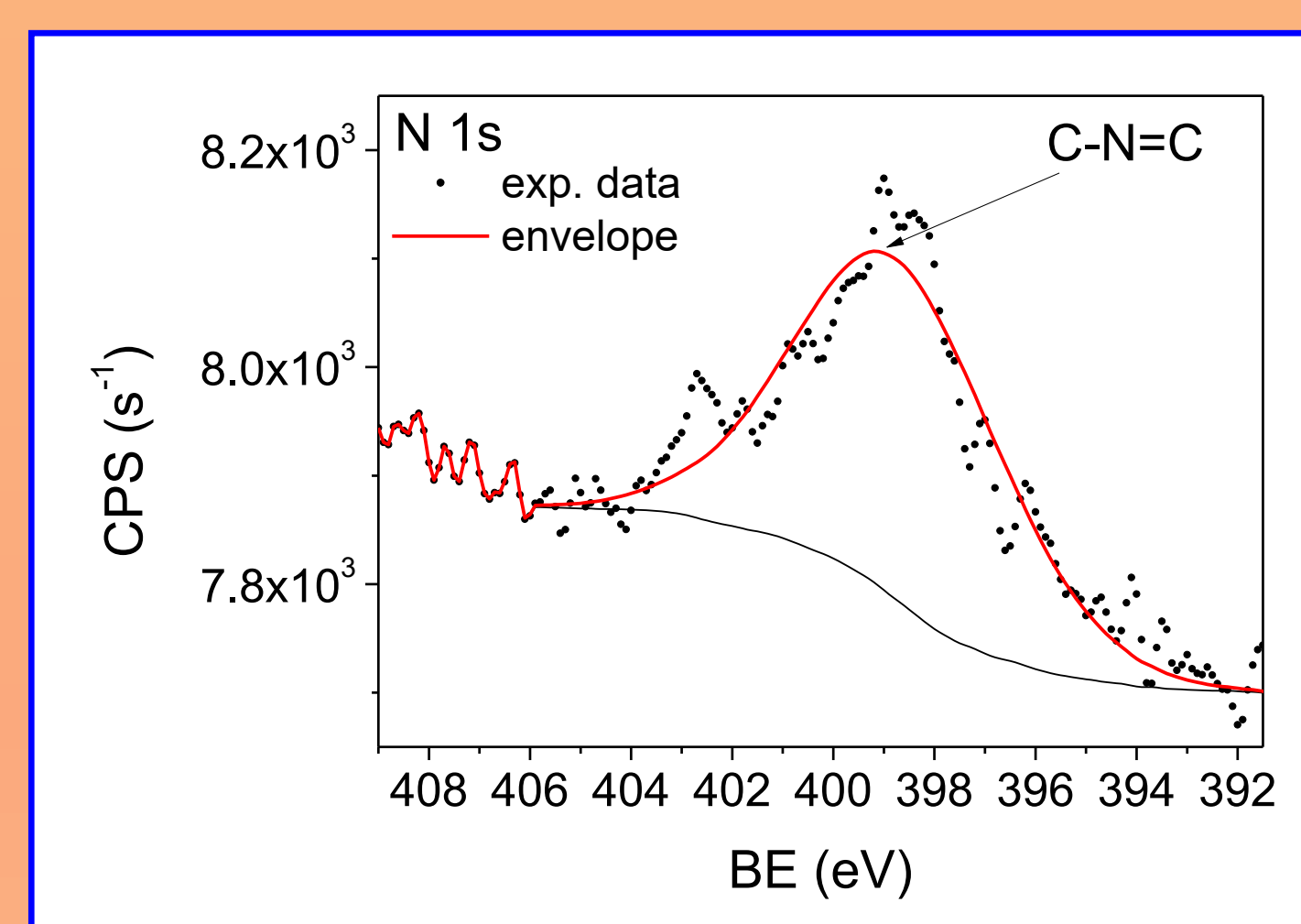
TEM image of g-C₃N₄-TiO₂:CuX% based nanocomposites



FT-IR spectra of g-C₃N₄-TiO₂:CuX% based nanocomposites



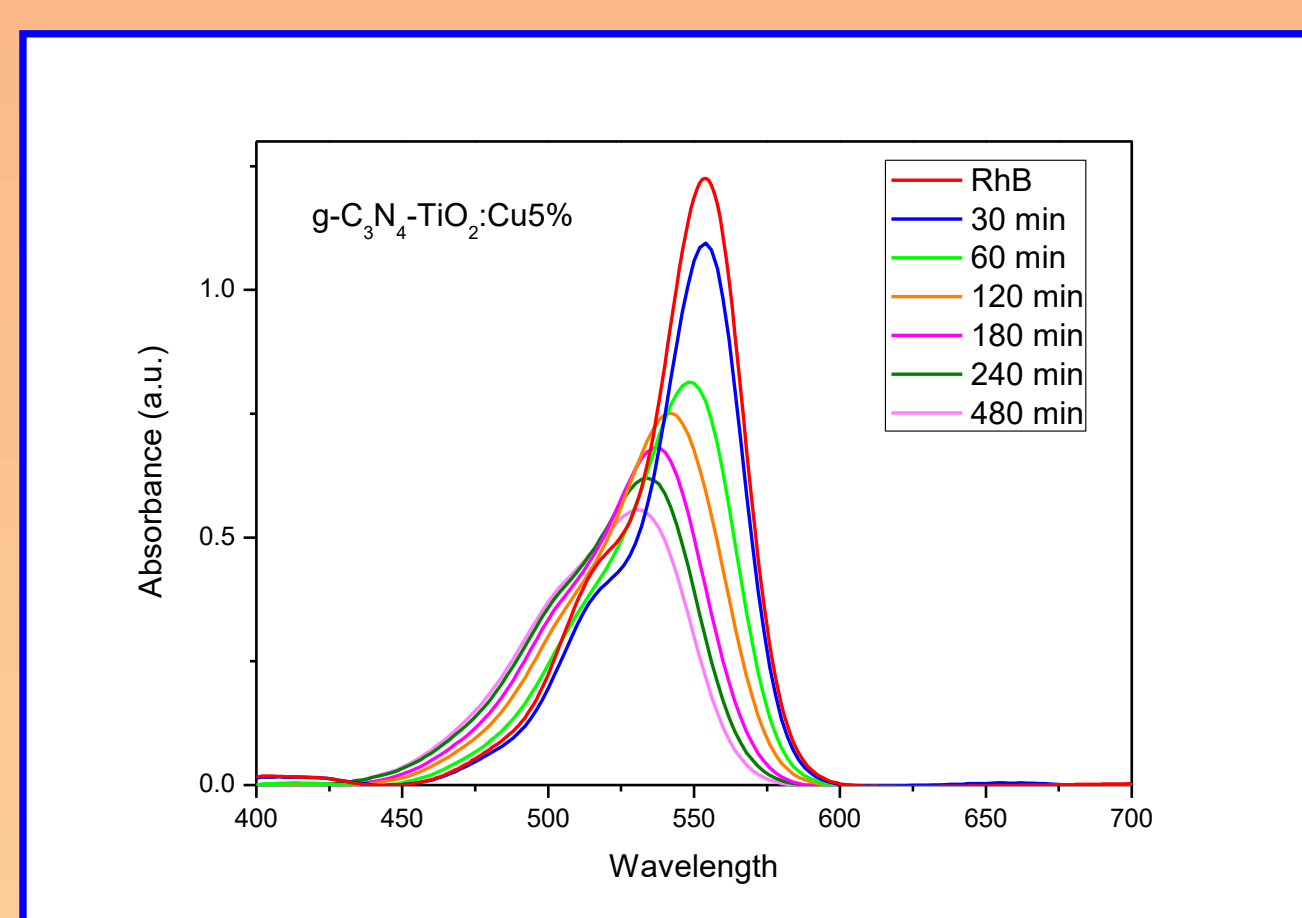
XPS survey spectrum of g-C₃N₄-TiO₂:Cu5% sample. Cu, Ti, O, C are identified in the spectrum.



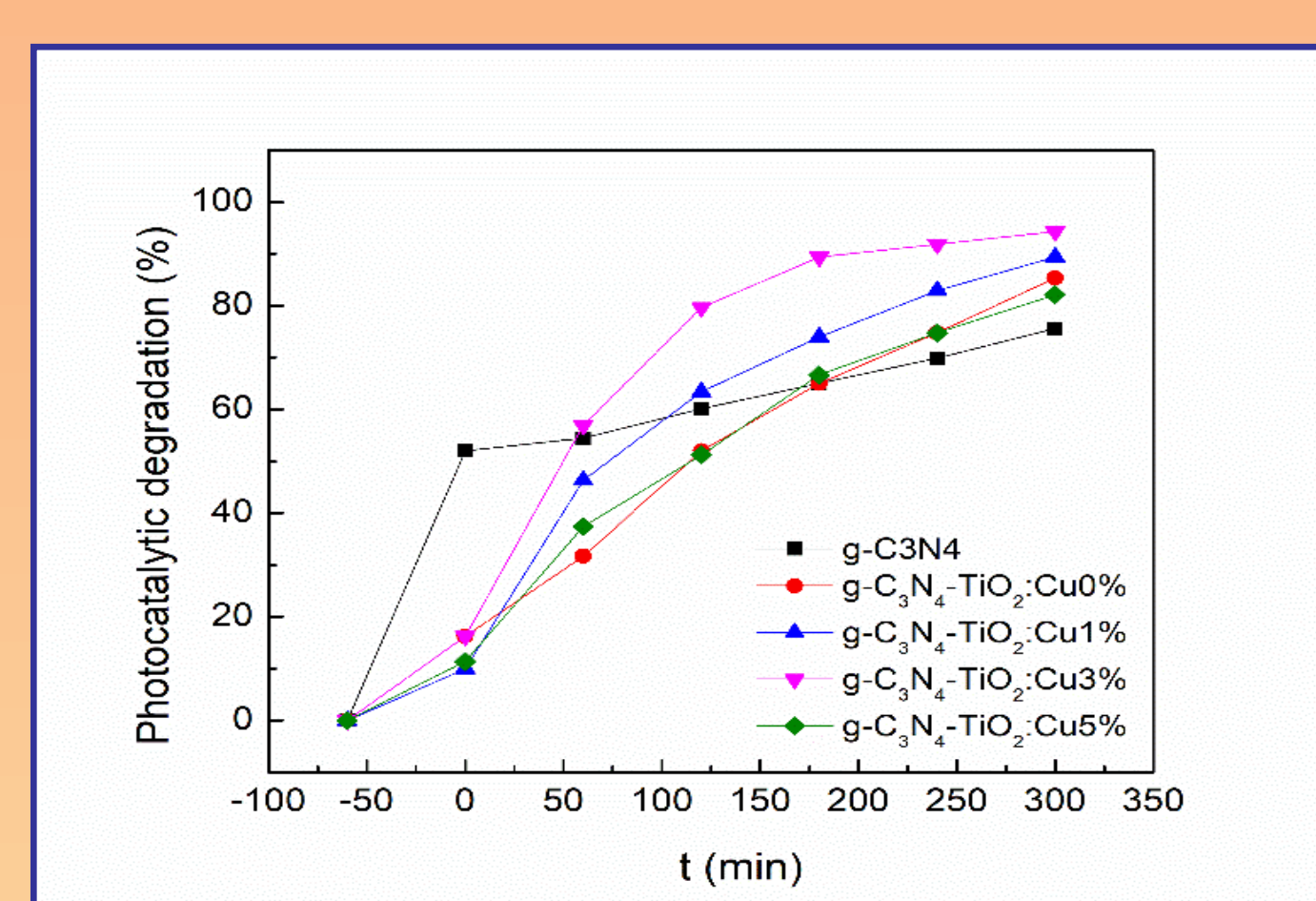
XPS spectrum of N1s recorded after Ar ion etching corresponding to 3.9nm depth. The observed peak at 397.5 eV is assigned to sp² bonded N atom in C-N=C triazine rings

Wavenumber [cm ⁻¹]	Bands Assignment
3438	stretching vibrations of O-H and N-H
2908	symmetric and asymmetric vibrations of C-H groups
2837	aromatic C-N stretching
2330	C=O stretching vibrations
1618	stretching or bending vibration of aromatic CN heterocycles including C-NH and C-N
1378	bending vibrations of C-H group from titanium acetylacetonate complex.
510	Ti-O stretching mode

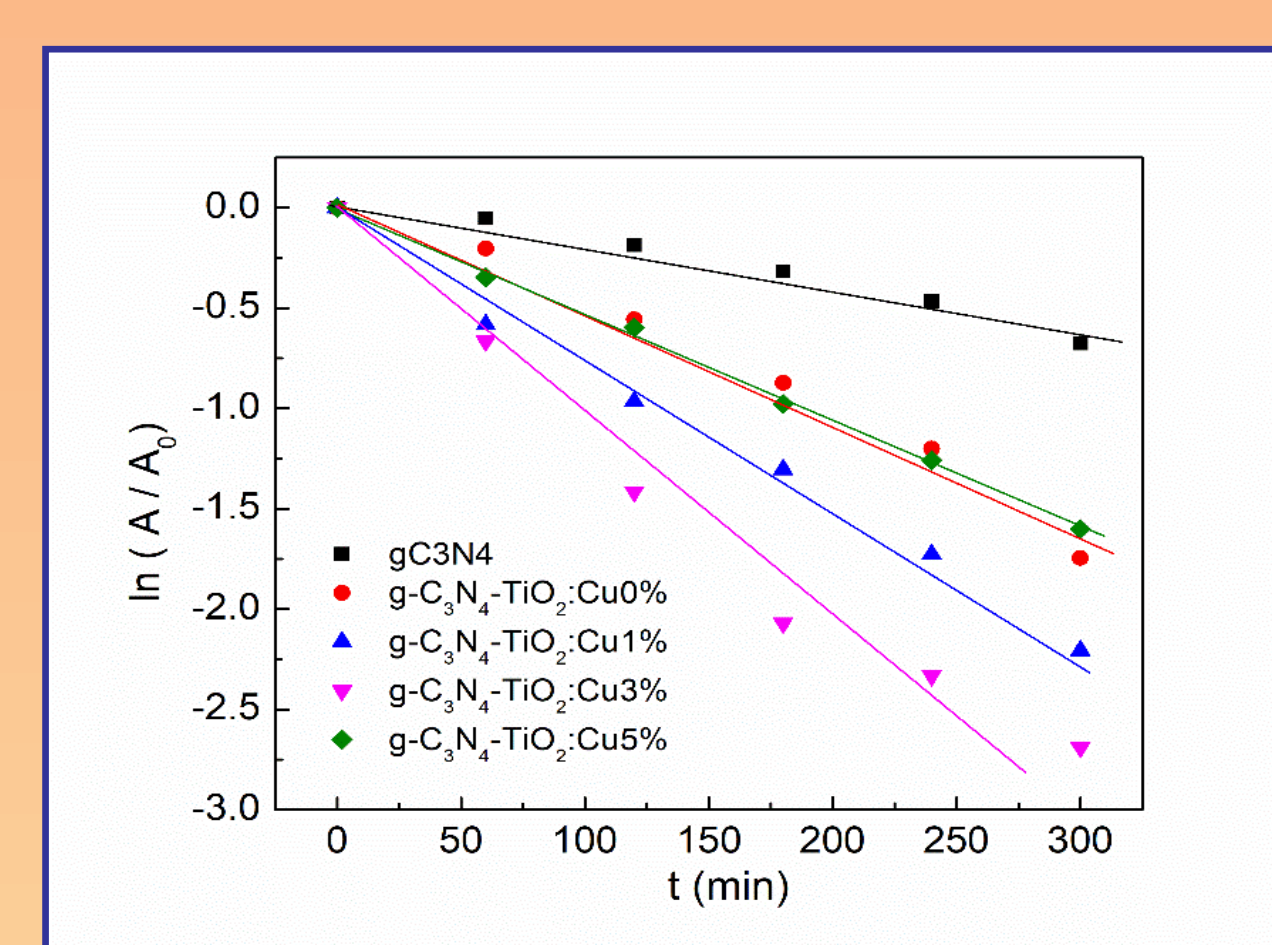
3.1 Photocatalytic properties



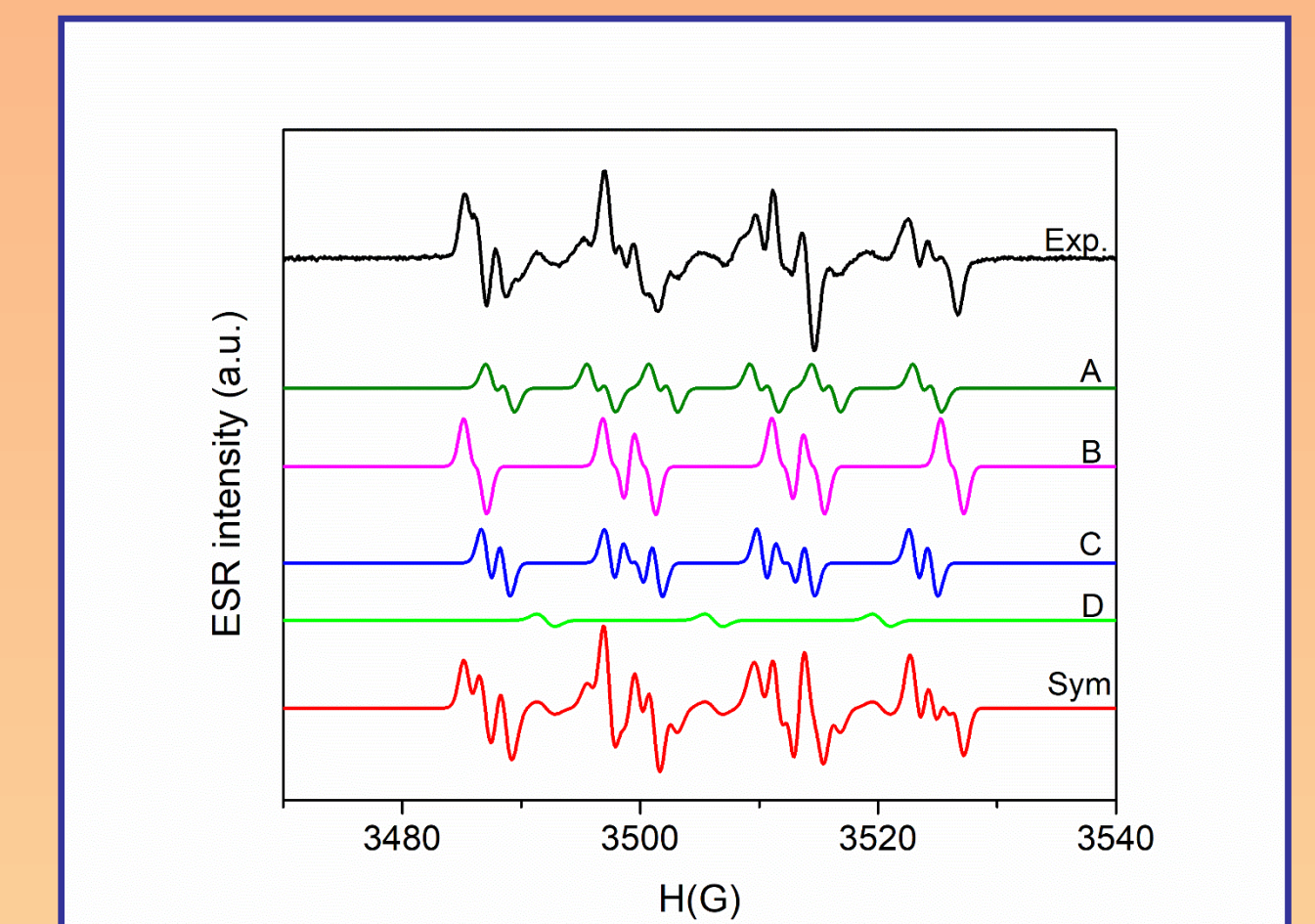
UV-Vis absorption spectra of RhB aqueous solution with g-C₃N₄-TiO₂:CuX% nanocomposites at different irradiation time intervals



Degradation curves of RhB in the presence of g-C₃N₄-TiO₂:CuX% samples



Evaluation of photodegradation kinetic



Experimental spectra and simulated spectra of DMPO spin adducts generated by g-C₃N₄-TiO₂:Cu3% sample. The simulated spectrum represents a linear combination of following spin adducts: ·DMPO-OCH₃ (a), ·DMPO-OOH (b), ·DMPO-O₂⁻ (c) and ·DMPO-N₂.

4. CONCLUSIONS

- g-C₃N₄ was synthesized by thermal degradation of urea, 2h at 550°C. After that Cu doped TiO₂ nanoparticles were growth on g-C₃N₄ by sol-gel alcoxidic route;
- XRD pattern illustrates the diffraction peaks corresponding to g-C₃N₄ nanoparticles and TiO₂ respectively.
- FT-IR spectra evidenced the specific stretching or bending vibration of aromatic CN heterocycles including C-N and C-NH. Also the specific vibrations corresponding to TiO₂ were observed.
- TEM images shows the morphology of samples. Compared to the pure g-C₃N₄ sample, TiO₂ agglomerates seem to be dispersed on the external surface of g-C₃N₄ in the case of composite samples.
- XPS spectra evidence qualitative composition of the samples. Cu, Ti, O, C and N are identified in the spectrum.
- All the samples show photocatalytic activity against RhB and the best performance was obtained using 3% Cu-doped TiO₂ based nanocomposites. The evidencing of Reactive oxygen species(ROS) were highlighted.
- From the simulation results that g-C₃N₄-TiO₂:Cu3% sample generates ROS species in almost equal quantity which are able to degrade the dye molecules.

ACKNOWLEDGEMENTS

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References

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