

C Varodi, F Pogacean, M Coros, V Mirel, L Magerusan, L-B Tudoran and S Pruneanu
National Institute for Research and Development of Isotopic and Molecular Technologies, 67-103 Donat Street, 400293 Cluj-Napoca, Romania

E-mail: codruta.varodi@itim-cj.ro

Introduction – 8-Hydroxy-2'-Deoxyguanosine Biomarker (8-OH-dG) is a cancer biomarker for tumors and it is used for the evaluation of oxidative damage processes in DNA. Graphite exfoliation with pulses of current has a lot of advantages for the graphene synthesis. In this work we prepared two graphene-based materials doped with nitrogen, sulfur and boron in different proportions, following denoted EXF-1, and EXF-2. Both materials were morphologically and structurally characterized by SEM, FTIR, and XRD and they were used for the modification of two glassy carbon electrodes, denoted GC/EXF-1 and GC/EXF-2. The electroanalytical parameters of the new electrodes were determined and tested in artificial plasma experiment with standard addition of 8-OHdG.

Experimental – The graphene samples were obtained and characterized by different techniques: SEM, X-Ray powder diffraction, FTIR, XPS, and elemental analysis. The electrodes were modified by drop-casting on their surfaces the graphene suspension in DMF and the electrochemical measurements were recorded with an AUTOLAB 302N electrochemical work-station (Metrohm-Autolab B.V., Netherlands). The Linear Sweep Voltammetry (LSV) measurements were generally run between 0.1 and 0.7 V vs Ag/AgCl, with a scan rate of 10 mV·s⁻¹ and chronoamperometric determinations were registered at +0.46 V.

Results and Discussions

Synthesis of graphene co-doped with heteroatoms: boron and nitrogen or sulfur

Table 1. Reaction conditions for graphene co-doped with heteroatoms: boron and nitrogen or sulfur (electrochemical) in pulsed current

| Probe | Electrolyte Conc. | Reaction time (min) | Applied Voltage (V) | Obs |
|-------|--|---------------------|---------------------|---|
| EXF-1 | 0,05M (NH ₄) ₂ SO ₄ +0,05M B(OH) ₃ + 0.05M NaCl | 240 | 12 | Pulsed current 0,4-0,8A |
| EXF-2 | 0,05M (NH ₄) ₂ SO ₄ +0,1M B(OH) ₃ + 0.05M NaCl | 270 | 12 | Pulsed current 0.50-0.55A, with tap water cooling |

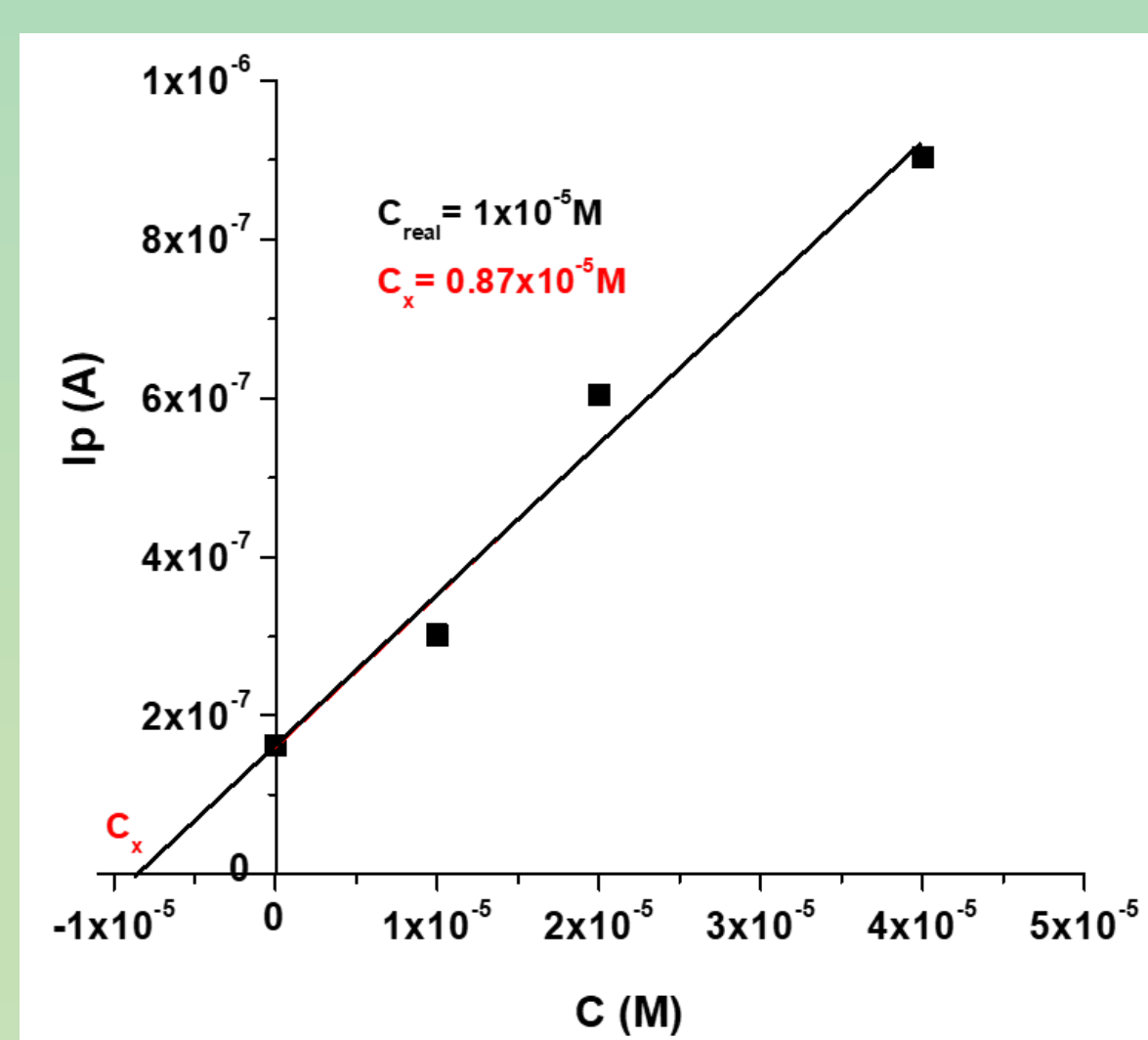
Structural characterization of graphene co-doped with heteroatoms: boron and nitrogen or sulfur

Table 2. Structural parameters obtained from X-ray diffractograms: crystallite size (D), interplanar distance (d), average number of layers (n) and quantity (%) of each type of graphene obtained

| Probe | 2θ (grade) | D (nm) | d (nm) | n | Obs |
|-------|------------|--------|--------|----|---------|
| EXF-1 | 9.401 | 2.342 | 0.951 | 2 | 19% GO |
| | 23.390 | 1.048 | 0.381 | 3 | 53% FLG |
| | 26.387 | 18.72 | 0.338 | 55 | 28% MLG |
| EXF-2 | 9.664 | 2.483 | 0.917 | 3 | 17% GO |
| | 21.685 | 1.181 | 0.410 | 3 | 69% FLG |
| | 26.224 | 10.103 | 0.341 | 30 | 14% MLG |

Standards addition of 8-OHdG in artificial plasma

Figure 3. Standard addition experiment at a potential of +0.46 V - calibration plot



Electrochemical detection of 8-OHdG

Figure 1. Linear voltammograms recorded with electrode GC/EXF-1 (A, C), GC/EXF-2 (B, D) modified with graphene electrochemically co-doped with heteroatom - boron and nitrogen or sulfur, in standard laboratory solutions (pH 6 PBS) containing increasing concentrations of 8-OHdG (10⁻⁷– 10⁻³ M). Calibration curves obtained for three distinct measurements.

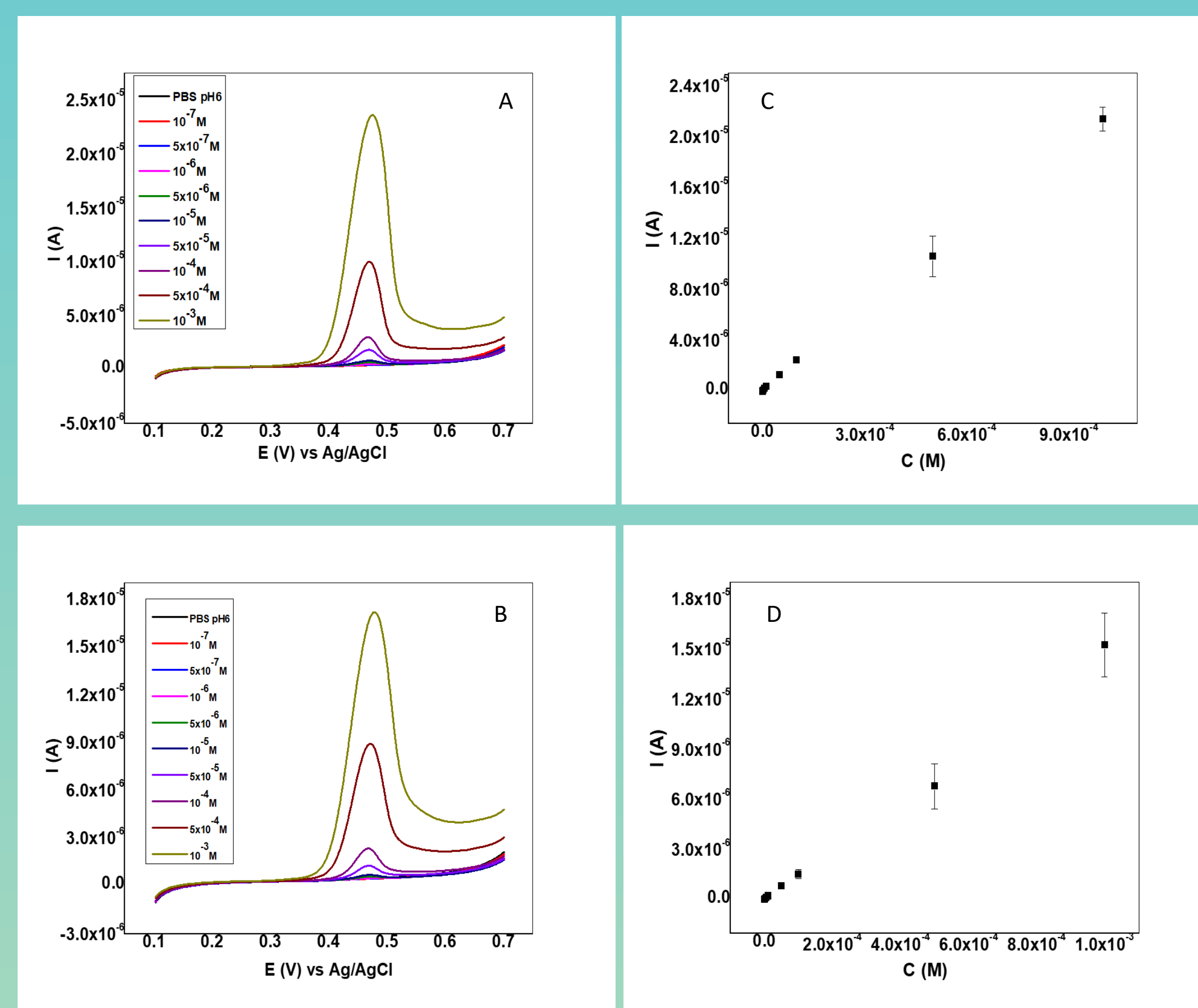
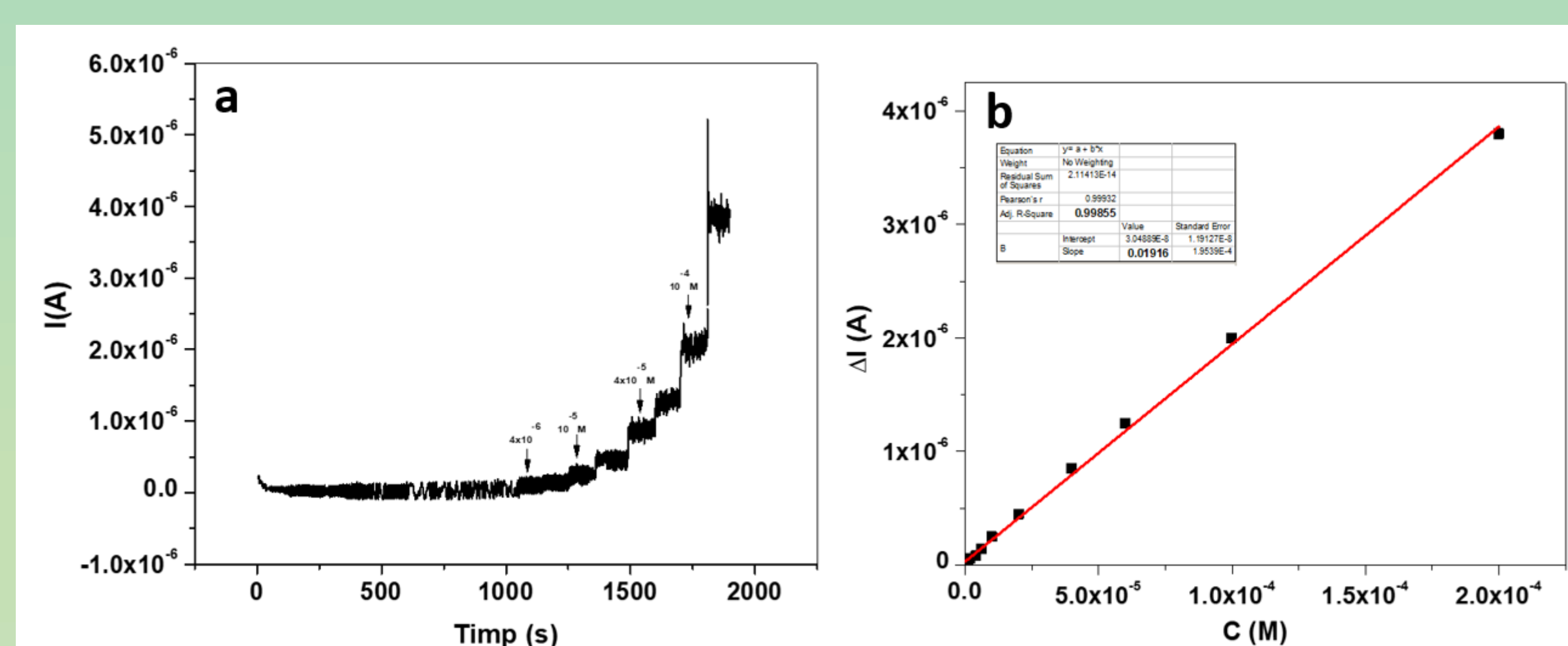


Figure 2. Amperometric curves recorded with GC/EXF-1 electrode in pH6 PBS supporting electrolyte, after the addition of known concentrations of 8-OHdG (a); The corresponding calibration plots (b).



Conclusions – There were prepared two graphene-based materials doped with nitrogen, sulfur and boron in different proportions, following denoted EXF-1, and EXF-2. Both materials were morphologically and structurally characterized and they were used for the modification of two glassy carbon electrodes, denoted GC/EXF-1 and GC/EXF-2. The electrochemical performances obtained for the 8-OH-dG biomarker detection with these new graphene-modified electrodes are very good (Detection limit of 0.9x 10⁻⁷ M, Linear range between 5x10⁻⁶– 10⁻³ M, sensitivity of 19 mA/M) and promising for real sample analysis.

Acknowledgement

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